

1 Calcification response of a key phytoplankton family to  
2 millennial-scale environmental change

3 H.L.O McClelland<sup>1</sup>, N. Barbarin<sup>2</sup>, L. Beaufort<sup>2</sup>, M. Hermoso<sup>1</sup>,  
4 P. Ferretti<sup>3,4</sup>, M. Greaves<sup>4</sup> & R.E.M. Rickaby<sup>1</sup>

- 4 1. Department of Earth Sciences, University of Oxford, South Parks Road, Oxford, OX1 3AN,  
5 UK
- 6 2. CEREGE CNRS-IRD-Aix Marseille Univ, Avenue Louis Philibert, BP80 13545 Aix en  
7 Provence cedex 04, France
- 8 3. Consiglio Nazionale delle Ricerche, Istituto per la Dinamica dei Processi Ambientali(CNR-  
9 IDPA), Calle, Larga Santa Marta 2137, Venice I-30123, Italy.
- 10 4. Godwin Laboratory for Palaeoclimate Research, Department of Earth Sciences, University  
11 of Cambridge, Downing Street, Cambridge CB2 3EQ, UK.

## 12 Abstract

13 Coccolithophores are single-celled photosynthesizing marine algae, responsible for half of the cal-  
14 cification in the surface ocean, and exert a strong influence on the distribution of carbon among  
15 global reservoirs, and thus Earth's climate. Calcification in the surface ocean decreases the  
16 buffering capacity of seawater for  $\text{CO}_2$ , whilst photosynthetic carbon fixation has the opposite  
17 effect. Experiments in culture have suggested that coccolithophore calcification decreases under  
18 high  $\text{CO}_2$  concentrations (  $[\text{CO}_2(\text{aq})]$  ) constituting a negative feedback. However, the extent  
19 to which these results are representative of natural populations, and of the response over more  
20 than a few hundred generations is unclear. Here we describe and apply a novel rationale for  
21 size-normalizing the mass of the calcite plates produced by the most abundant family of coccol-  
22 ithophores, the Noëlaerhabdaceae. On average, ancient populations subjected to coupled gradual  
23 increases in  $[\text{CO}_2(\text{aq})]$  and temperature over a few million generations in a natural environment  
24 become relatively more highly calcified, implying a positive climatic feedback. We hypothesize  
25 that this the result of selection manifest in natural populations over millennial timescales, so has  
26 necessarily eluded laboratory experiments.

## 27 Introduction

28 Coccolithophores are the modern ocean's dominant calcifying phytoplankton and coccoliths, the  
29 distinctive calcite plates that they produce, have populated the fossil record for over 200 million  
30 years[1, 2]. Biogenic calcification is an important climatic feedback[3, 4], and in most organ-  
31 isms is thought to be strongly affected by changes in the carbonate chemistry of seawater. In  
32 coccolithophores, calcite precipitation occurs inside the cell, so the site of calcite precipitation  
33 is buffered from the external environment and is subject to an unusually high degree of bio-  
34 logical control. The most abundant modern species of coccolithophore are *Emiliana huxleyi*  
35 and *Gephyrocapsa oceanica*; bloom-forming members of the family Noëlaerhabdaceae (Hapto-  
36 phyta: Coccolithophyceae: Isochrysidales: Noëlaerhabdaceae). Laboratory culture experiments  
37 have shown that these two species tend to calcify less when carbon dioxide concentrations are  
38 higher[5, 6, 7], although the trade off between DIC availability and pH appears to be critical  
39 [8]. Yet these observed responses each characterize a single genome, adapted to the modern  
40 environment and exposed to artificially manipulated conditions in the laboratory in the absence  
41 of genetic change (phenotypic plasticity). In nature, communities are genetically and phenotypi-  
42 cally heterogeneous, and selective pressures lead to competition. Environments are dynamic, and  
43 changes in average conditions are subtle and prolonged. Sexual reproduction, which has never  
44 been observed in the laboratory in coccolithophores, genetically homogenizes a population within  
45 a biological species, whilst accelerating adaptation through propagation of the most beneficial  
46 combinations of alleles and removal of deleterious mutations via natural selection. To eluci-  
47 date the role of these phytoplanktonic marine calcifiers in nature across geological history, their  
48 response must be studied on timescales of tens to tens of thousands of years, within natural pop-  
49 ulations, where evolutionarily relevant processes such as meiosis and syngamy may occur[9, 10],  
50 and where the rate and amplitude of environmental change is representative of the real world. In-  
51 deed a theoretical model has suggested that on such timescales the selective pressure of increasing  
52  $[\text{CO}_2(\text{aq})]$  may have favoured the more heavily calcified forms of coccolithophore[11].

53 An apparent dichotomy exists between the consensus view of phenotypic plasticity as ob-

## Figure 1

54 served in short-term experiments and the theoretical result of long-term evolutionary adaptation.  
55 However, experiments lasting hundreds of generations have shown that asexual coccolithophore  
56 populations have the potential to adapt in culture[12, 13] and large-scale surveys have revealed  
57 trends across spatial environmental gradients in nature[14, 15]. The fossil record, by contrast,  
58 is an archive of information about ancient natural coccolithophore communities that responded  
59 to real environmental changes over geological timescales. The challenge is to extract meaningful  
60 information from this resource. Isolated coccoliths in deep-sea sediment are often the only rem-  
61 nants of ancient coccolithophores to survive geological time, so inferences about the physiology  
62 of coccolithophores that lived in the past must come from this evidence alone.

## 63 Results

### 64 Size-normalization of coccolith mass

65 Studies to date have yielded contradictory results regarding the response of calcification to en-  
66 vironmental change on geological timescales [16, 14], but this disagreement is, at least in part,  
67 a result of the lack of consistency between the parameter measured, and how this is inferred to  
68 represent "calcification intensity". Coccolith mass has been used extensively to infer calcification  
69 ability[16, 17, 14, 15, 18, 19], but as it is not independent of coccolith size, size-normalization is  
70 highly nontrivial and it remains a biologically abstract quantity. In order to solve this problem,  
71 we have developed a procedure for size-normalising coccolith mass, by correlating an index based  
72 on coccolith morphometry with the molar ratio of particulate inorganic to particulate organic  
73 carbon (PIC:POC) of the biomass. The PIC:POC ratio is a direct record of the ratio of inte-  
74 grated rates of calcification to photosynthesis[3], and therefore describes an energetic and carbon  
75 budget trade off between calcification and biological (metabolic) activity that makes more sense  
76 biologically than coccolith mass in the context of physiology and adaptation, and is indepen-

77 dent of coccolithophore size. Through size-normalising mass by consideration of a dimensionless  
78 but biologically meaningful parameter, we circumvent problems associated with allometry. The  
79 concept for size-normalising coccolith mass is based on the following arguments:

80 (i) The molar PIC:POC ratio of coccolithophore biomass is proportional to the ratio of spher-  
81 ical volumes of calcite to organic matter (Fig.1B).

82 (ii) The square root of coccolith area ( $\sqrt{A_L}$ ), which we use as a 1-dimensional measure of  
83 coccolith size, is proportional to coccosphere radius ( $R_s$ ):  $R_s \propto \sqrt{A_L}$  (Fig.1C).

84 (iii) Coccolith thickness ( $T_L$ ) is proportional to coccosphere thickness ( $T_s = R_s - R_c$ ):  $T_s \propto T_L$   
85 (Fig.1D).

86 (iv) Given (i), (ii) and (iii), a combination of coccolith mass and area can be related back to  
87 coccolithophore PIC:POC, via cell and coccosphere dimensions, thus providing a rationale  
88 for size-normalisation of mass (Fig.1E).

89 We tested each assumption and calibrated our mass-normalisation concept using cultured  
90 coccolithophores. We grew two strains each of *E. huxleyi* and *G. oceanica* in artificial seawater  
91 adapted for four different DIC concentrations at constant pH, temperature and nutrient levels,  
92 in duplicate (see methods). (i) was established via *in vivo* measurement of coccosphere and cell  
93 volumes, and direct measurement of the molar PIC:POC ratio of the biomass:

$$\ln(PIC : POC_{volume}) \propto \ln\left(\frac{V_s - V_c}{V_c}\right) = \ln\left(\left(\frac{R_s}{R_c}\right)^3 - 1\right), \quad (1)$$

94 where  $V_s$  and  $V_c$  are respectively the volumes of the coccosphere and the cell, and  $R_s$  and  $R_c$  are  
95 respectively the radii of the coccosphere and the cell (as labelled in Fig.1A). Eq. 1 is shown in  
96 Fig. 1B.

97 To generate disarticulated coccoliths, the filtered culture residue was bleached to remove  
98 organic matter whilst preserving the calcite intact, emulating the effect of decomposition. We  
99 analyzed this synthetic coccolith-sediment using SYRACO[20]; an automated image analysis  
100 tool, which measures coccolith area and mass (see methods). Our data support assumptions (ii)

101 and (iii) (Fig.1C and Fig.1D respectively). The relationship that we find between coccolith size  
 102 and cell size agrees with the well established relationship between coccolith length and cell size  
 103 in the Noëlaerhabdaceae[21], which is derived from fossils (red dashed line Fig.1C). That our  
 104 strains all lie on the same line as fossil Noëlaerhabdaceae supports the theory that this family of  
 105 coccolithophores are a continuum of forms[21, 14]. Eq.1 can be simplified by approximating the  
 106 cell and coccosphere as having the same radius ( $R_s$ ). The volume of organic material is then the  
 107 volume of a sphere of radius,  $R_s$ , and the volume of calcite in the coccosphere is that of a thin  
 108 shell of thickness  $T_s$  and radius  $R_s$ . Using assumptions (ii) and (iii) gives:

$$\begin{aligned}
 \ln(PIC : POC_{coccolith}) &\propto \ln\left(\frac{R_s^2 \times T_s}{R_s^3}\right) = \ln\left(\frac{T_s}{R_s}\right) \\
 &\propto \ln\left(\frac{T_L}{\sqrt{A_L}}\right) = \ln\left(\frac{M_L}{A_L^{3/2}}\right), \tag{2}
 \end{aligned}$$

110 where  $M_L$  = coccolith mass, and  $A_L$  = coccolith area are the direct output from the SYRACO  
 111 analysis, and  $M_L/A_L = T_L$ . As per (iv), an ordinary least squares linear regression incorporating  
 112 an estimate of the uncertainty in PIC:POC based on the  $1\sigma$  prediction interval gives:

$$PIC : POC_{coccolith} = e^{3.5 \pm 0.2} AR_L^{1.12}, \tag{3}$$

113 where  $AR_L = \frac{T_L}{\sqrt{A_L}} = M_L/A_L^{3/2}$  is the lateral cross-sectional aspect ratio of a coccolith ( $AR_L$ ).

114 Equation 3 is derived from first principles, and is supported empirically (Fig.1.E).  $AR_L$  is  
 115 taken to be the most sensible way of size-normalising coccolith mass, which alone is biologically  
 116 meaningless. Caveats associated with directly translating  $AR_L$  to PIC:POC using Eq.3 are  
 117 discussed later.

## 118 Down-core experiment

119 To explore the calcification response of natural coccolithophore populations to climatic change in  
 120 the form of a down-core experiment, we used this novel approach to analyze sediment-core mate-  
 121 rial from two glacial terminations, which are of contrasting magnitude (Fig.2). The larger of these

122 terminations sees an increase in atmospheric CO<sub>2</sub> mixing ratios (CO<sub>2atm</sub>) similar in magnitude  
123 to that since the industrial revolution, albeit two orders of magnitude slower. SYRACO analysis  
124 was performed on sediment from ODP site 1123, on Chatham rise, east of New Zealand in the  
125 southernmost Pacific (41°47.2'S, 171°29.9'W, 3290m water depth). Depths span the penultimate  
126 and 6th most recent glacial-interglacial cycles (MIS 7-5:~200-100ka and MIS 15-13:~600-500ka,  
127 respectively), and sample resolution is higher (~1000 year) over the terminations (TII:~140-  
128 125ka and TVI:~540-520ka). Dissolution proxies[22] and SEM observations of coccoliths show  
129 good preservation at this site throughout the periods of interest. ODP site 1123 exists at the  
130 northern edge of the sub tropical front (STF) in the southernmost Pacific Ocean, and may  
131 see an increased influence of sub antarctic water during glacials and subtropical water during  
132 interglacials[22]. We used *in situ* proxy reconstructions of sea surface temperature (SST) and  
133 relative nutrient concentration to infer surface ocean conditions throughout the periods of in-  
134 terest. We calculated [CO<sub>2</sub>(aq)] from published CO<sub>2atm</sub> records from ice cores (see methods)  
135 and from SST inferred from planktic foram Mg/Ca ratios (see methods), assuming equilibrium  
136 between the atmosphere and surface waters. [CO<sub>2</sub>(aq)] and temperature increased during each  
137 glacial termination, with no obvious change in relative nutrient concentrations (see methods and  
138 Fig.S2). pH and [DIC] were unconstrained, but on these timescales, the decrease in pH due  
139 to invasion of CO<sub>2</sub> may be partially buffered by dissolution of carbonate sediments. For this  
140 reason, [CO<sub>2</sub>(aq)] is used to describe the carbonate system, as this is the most well constrained  
141 component.

142 SYRACO analysis[20] measures the mass and area of each coccolith belonging to the family  
143 Noëlaerhabdaceae in the analysed samples. AR<sub>L</sub> is calculated subsequently. Additionally, each  
144 coccolith is assigned to a morphotaxonomic group on the basis of shape. These groupings approx-  
145 imate species-level taxonomic classifications and provide insight into subsets of the population.  
146 Here we show morphotaxon-specific records for the groups dominated by large *Gephyrocapsa* spp.  
147 and by *E. huxleyi* (MG<sub>Geo</sub> and MG<sub>Emi</sub> respectively). These species are the most thoroughly  
148 studied in the laboratory, and their ultrastructure is suited to SYRACO analysis.

Figure 2

## 149 Down-core results

150 Over TII, mean Noëlaerhabdaceae coccolith mass and area increase transiently across the ter-  
151 mination. This is accompanied by an increase in the mean-independent variance (arithmetic  
152 coefficient of variation; ACV) in both of these variables (Fig.2), implying a non-uniform re-  
153 sponse within the population. Specifically, the mass and area of  $MG_{Geo}$  transiently increase in  
154 parallel (almost doubling) over the termination, but ( $MG_{Emi}$ ) shows a subtle decrease. Over  
155 TVI, prior to the first appearance of *E.huxleyi* coccoliths in the fossil record at around 290ka[23],  
156 the absolute magnitude and ACV of Noëlaerhabdaceae coccolith mass is very stable, though with  
157 a large decrease in ACV during the MIS14 glacial inception. The absolute magnitude of coc-  
158 colith area, however, peaks during MIS14 glacial maximum and decreases during TVI, but with  
159 little change in the area ACV. The time series of population mean  $AR_L$  (size-normalised mass)  
160 exhibits an increase across both terminations. This effect is paralleled but larger in the  $AR_L$   
161 of  $MG_{Geo}$ , which increases by  $\sim 30\%$  across both TII and TVI. By contrast, the  $AR_L$  response  
162 of  $MG_{Emi}$  across TII is of a slight decrease. The absence of *E. huxleyi* explains the higher  
163 population mean  $AR_L$  over TVI than over TII, and its presence and opposing response to *G.*  
164 *oceanica*, as inferred from the morphotaxa, explains the increased variance observed over TII.  
165 The parameter  $AR_L$  reconciles the differing response of mass and area of  $MG_{Geo}$  to increasing  
166  $[CO_2(aq)]$  and temperature across these two terminations.

## 167 Discussion

168 The  $AR_L$  time series implies that coccoliths become more calcified in Noëlaerhabdaceae, and  
169 especially *G.oceanica*, with increasing  $[CO_2(aq)]$  and temperature at site 1123 in natural pop-  
170 ulations over a timescale of thousands of years. This increase contrasts with results from the  
171 literature that describe the capacity for phenotypic plasticity in these organisms. In the labora-



172 tory, single strains of *G. oceanica* calcify relatively less with increasing  $[\text{CO}_2(\text{aq})]$  [7, 6, 24, 5],  
173 and there is no discernible effect of temperature[7]. *E. huxleyi*'s plastic response to increasing  
174  $\text{CO}_2$  concentration over the reconstructed  $[\text{CO}_2(\text{aq})]$  range of our time series is of decreasing  
175 calcification when [DIC] or alkalinity (ALK) is held constant[25, 7, 5, 6, 26] but the opposite at  
176 constant pH[26]. In *E. huxleyi* the plastic effect of temperature in the laboratory is large, but  
177 non-linear and poorly constrained[7].

178 So far, our proposed method for size-normalising coccolith mass has used the PIC:POC ratio  
179 simply to justify our theoretical rationale with real biological data. It is however, possible to  
180 use this relationship as a proxy for PIC:POC, using the coccolith aspect ratio as an input. The  
181 predictive power of Eq.3 as a proxy for Noëlaerhabdaceae PIC:POC is dictated by the spread  
182 of the residuals about the regression line: the prediction interval - not only by the confidence  
183 interval, which describes the uncertainty in the relationship[27]. With due consideration of the  
184 associated caveats, and formal consideration of uncertainty, Eq.3 can be used as a proxy to di-  
185 rectly infer PIC:POC from Noëlaerhabdaceae coccolith morphometry (Fig.3). If the constant of  
186 proportionality between with size and cell size is systematically influenced by the environment or  
187 differences between species, this will be reflected in the estimate of PIC:POC. In our calibration,  
188 hundreds of thousands of cells, and thousands of individual coccoliths were measured for each  
189 sample, which has the effect of averaging over the potential variation in the slope of Eq.3 due  
190 to variations in the number of coccoliths per cell, or the degree of overlap of coccoliths. Indeed,  
191 the established relationship between coccolith size and cell size in heterogeneous natural fossil  
192 populations [21] agrees with ours very well (Fig.1C), lending support to the constant relationship  
193 between cell size and coccolith size, and application to the whole Noëlaerhabdaceae family. Coc-  
194 colith over-production in *E.huxleyi*, the only species where multiple coccolith layers are produced,  
195 will cause coccolith morphometry to underestimate net PIC:POC, and this is unavoidable. In  
196 the rare deposits where preservation is exceptional and coccospheres are intact, such as in the  
197 Lagerstätte deposits of Tanzania[28, 29], Eq.1 may be used to estimate PIC:POC directly. This  
198 approach would bypass the assumptions necessary to infer PIC:POC directly from disarticulated

### Figure 3

199 coccoliths.

200 With the aforementioned caveats in mind, changes in  $AR_L$  can be used to directly estimate  
201 the PIC:POC ratio of the biomass in these ancient organisms using Eq.3, albeit with a rather  
202 large associated uncertainty. We have compiled PIC:POC values from the literature and com-  
203 pared them with results inferred from our down-core record (Fig.3 and Fig.S1). The PIC:POC of  
204 both the Noëlaerhabdaceae family mean, and of  $MG_{Geo}$ , increases in response to  $[CO_2(aq)]$  and  
205 temperature. This is opposite to results from the laboratory (Fig.3). Any change in PIC:POC  
206 inferred from  $AR_L$  of  $MG_{Emi}$  coccoliths are well within the range of uncertainty however. Al-  
207 though the uncertainties in these estimates are large, these results are necessarily elusive to  
208 laboratory experiments due to the timescales involved.

209 Phenotypic plasticity, as observed in laboratory experiments to date, cannot explain coccoliths  
210 becoming more calcified with increasing  $[CO_2(aq)]$  and temperature. Based on our down-core  
211 observations we infer, therefore, either a component of selection for more heavily calcifying forms,  
212 or that the parameter space, and thus phenotypic plasticity, has not been thoroughly explored  
213 in the laboratory. Selection may drive evolution of the population internally through differential  
214 suitability of forms to the abiotic environment, or through differential susceptibility to grazers or  
215 viral attacks [30], or via propagation of differentially adapted forms introduced from elsewhere  
216 along gradients of nutrients, pH, salinity or temperature. Alternatively the apparent increase in  
217 size-normalised coccolith mass, within  $MG_{Geo}$  in particular, may be due to a transient increase in  
218 the relative abundance of established more highly calcifying pseudo-cryptic sub lineages[14, 31].  
219 The increased mean-independent variance in size-normalised coccolith mass across TII may be the  
220 result of selection for different growth strategies, and/or differences in phenotypic plasticity across  
221 the population. As a substrate, low  $CO_2$  concentrations may limit growth rate [32], so increasing  
222 its concentration may allow for faster growth; but elevated  $CO_2$  at constant alkalinity decreases  
223 the saturation state of calcite making it more energetically costly to calcify. Increases in  $CO_{2atm}$

#### Figure 4

224 also drive increases in temperature. To explain the increased variance over TII by the differential  
225 effect of phenotypic plasticity within the population superimposed on a general selection for more  
226 highly calcified forms, the more lightly calcifying forms would need to be more susceptible to a  
227 decreasing saturation state, but this is not observed in culture. If anything, the relatively heavily  
228 calcifying *G. oceanica* appears to exhibit a greater plastic decrease in size-normalised coccolith  
229 mass with increasing  $[\text{CO}_2(\text{aq})]$  than the relatively lightly calcifying *E. huxleyi*. Alternatively,  
230 increasing  $[\text{CO}_2(\text{aq})]$  could conceivably create diverging niches; lightly calcifying forms could be  
231 selected on the basis of faster growth on a given the  $\text{CO}_2$  substrate, and highly calcifying forms  
232 could be selected on the basis of reduced mortality due to the greater integrity of their calcitic  
233 (presumably defensive) structure. Whatever the mechanism, it is interesting to note that the  
234 size-normalised coccolith mass of the population is driven to a very narrow range during glacial  
235 times (Fig.2). If this observation does indeed reflect a narrow range of PIC:POC values, it may  
236 indicate that it is universally energetically favourable, at low temperature and  $[\text{CO}_2(\text{aq})]$ , to  
237 maintain a balance between the relative rate of photosynthesis to calcification, thus minimizing  
238 the effect of biomass production on intracellular carbonate chemistry (Fig.4).

239 From a biogeochemical viewpoint, calcification and photosynthesis alter the buffering capac-  
240 ity of water for  $\text{CO}_2$  in opposing ways through their respective effects on  $[\text{DIC}]$  and  $[\text{ALK}]$ . These  
241 effects are relevant on scales ranging from the sub-cellular environment in coccolithophores to  
242 global biogeochemical cycles. In the sub-cellular environment, the ratio of calcification to pho-  
243 tosynthesis determines the steady state drift in intracellular pH, that would otherwise occur if  
244 it were not counteracted by the active pumping of protons into or out of the cell (Fig.4). Dur-  
245 ing blooms, coccolithophores constitute an instantaneous net sink or source of  $\text{CO}_2$ , or drive  
246 a change in pH, depending on their PIC:POC ratio (Fig.4). Under typical modern conditions,  
247 formation of biomass in the surface ocean with a PIC:POC of higher or lower than  $\sim 1.86$  consti-  
248 tutes respectively an instantaneous source or sink of  $\text{CO}_2$  (Fig.4), with this critical value varying

249 with the carbonate chemistry of the water. This effect is not permanent however; the effect of  
250 biomass production on surface ocean seawater chemistry is often transient due to respiration and  
251 remineralization of organic matter by grazers, and dissolution of calcite causing cycling within  
252 the surface ocean. Carbon fluxes between the atmosphere and ocean are strongly influenced by  
253 export of organic matter and calcite from the surface ocean. Export has a prolonged effect on  
254 surface ocean/atmosphere partitioning of CO<sub>2</sub> because DIC and ALK are sequestered in the deep  
255 ocean or sediments, which have residence times of respectively hundreds, and tens of thousands  
256 of years. In constituting up to half of the calcification in the surface ocean [1], of which members  
257 of the family Noëlaerhabdaceae typically contributing on the order of half [33], it is through their  
258 role in pumping calcite to depth that coccolithophores constitute a significant lever on the global  
259 climate system.

260 In the modern ocean, phytoplankton stock is a function of light, nutrients, grazing and viral  
261 lysis [34]. The effect of predators and viruses cannot explain the current distribution of primary  
262 production in the surface ocean, which is dominated by nitrate and iron availability [35]. In  
263 coccolithophores, the PIC:POC ratio describes the amount of calcite produced per unit organic  
264 matter. As organic matter production is finite and limited by the availability of biologically  
265 accessible nitrogen and iron, the PIC:POC ratio directly corresponds to the amount of calcite  
266 produced for a given standing stock or flux of ultimately limiting nutrients. In the absence of  
267 other changes, an increase in the PIC:POC ratio of coccolithophores in the surface ocean leads  
268 to an increase in the ratio of ALK:DIC exported, and thus a decrease in the buffering capacity  
269 of the surface ocean for CO<sub>2</sub>. A confounding effect arises because cells are approximately the  
270 same density as water, and must be ballasted by heavy minerals, including calcite, in the form  
271 of aggregates or faecal pellets in order to be exported from the surface ocean. In terms of the  
272 net effect on seawater carbonate chemistry in the surface ocean, export of calcite may therefore  
273 constitute a trade-off between direct removal of calcite, and the effect this has on the rate of  
274 export of organic matter [36].

275 In our down core record, coccolithophores belonging to the family Noëlaerhabdaceae appear,

276 on average, to calcify more under increasing  $[\text{CO}_2](\text{aq})$  and temperature. Although it is impos-  
277 sible to decouple the effects of temperature,  $[\text{CO}_2(\text{aq})]$ , salinity and nutrient availability down-  
278 core, these parameters have varied together throughout geological time. On glacial-interglacial  
279 timescales therefore, to first order, the Noëlaerhabdaceae' may actually constitute a positive  
280 feedback to increasing  $\text{CO}_{2atm}$  on millennial timescales.

281 We have shown that natural coccolithophore populations appear to adapt to rising  $[\text{CO}_2(\text{aq})]$   
282 and temperature on a millennial timescale, dominantly via selection for an increased tendency  
283 to calcify. Thus, this work introduces a temporal dimension to the prevailing view based on  
284 the results of culture manipulation experiments. The theoretical model predicting this outcome  
285 describes a trade-off between fast growth and calcification[11], confounding the implications for  
286 the total rate of calcite production in the surface ocean. We anticipate our results to be of use  
287 to biogeochemical modellers, but a more thorough understanding of the fate of biogenic material  
288 produced in the surface ocean is essential before the full implications of this work are realised.

## 289 Methods

### 290 Culture experiments.

291 Duplicate monoclonal batch cultures of four strains of coccolithophore belonging to the family  
292 Noëlaerhabdaceae were grown in sterile filtered ( $0.2\mu\text{m}$ ) artificial seawater prepared according to  
293 ESAW[37] adapted for a range of DIC concentrations ( $[\text{DIC}] = 1.380\text{mM}, 2.147\text{mM}, 3.067\text{mM}$  and  
294  $6.135\text{mM}$ ) at constant pH (8.2) by varying sodium bicarbonate addition and titration with HCl  
295 and with nitrate ( $442\mu\text{M}$ ), phosphate ( $5.00\mu\text{M}$ ), vitamins, trace metals and EDTA according  
296 to K/2[38]. Carbonate chemistry manipulation at constant pH is more analogous to changes  
297 expected in the surface ocean on a glacial-interglacial timescale than holding alkalinity constant,  
298 due to buffering by carbonate sediments. Cultures were maintained at  $15^\circ\text{C}$  with an incident  
299 photon flux of  $250\mu\text{E}$  and a 12/12 light/dark cycle. Cells were acclimated for  $>20$  generations in  
300 dilute batch culture for each experimental condition prior to inoculation. Cells were inoculated  
301 in 2.4l polycarbonate flasks, with no headspace and sealed off to the air with teflon lined caps.  
302 Removal of medium during the experiment was unavoidable due to the need to count and measure  
303 cells, and resulted in a maximum headspace of  $20\text{cm}^3$  at harvest. In order to minimise the  
304 drift in culture conditions throughout the course of the experiment, cells were harvested at  
305  $\sim 1\text{-}2\%$  (and never greater than 4%) of maximum cell density, which was determined for each  
306 experimental condition and strain combination via preliminary experimentation. Strains were  
307 AC478 (RCC1211 *Gephyrocapsa oceanica* from Portuguese coast in Atlantic Ocean), AC472  
308 (RCC1216 *Emiliana huxleyi*, from Tasman Sea in Pacific Ocean), AC448 (RCC1256 *Emiliana*  
309 *huxleyi*, Icelandic coast in Atlantic Ocean) and AC279 (RCC1314 *Gephyrocapsa oceanica*, French  
310 coast in Atlantic Ocean) from the Roscoff culture collection (RCC). Particulate material was  
311 harvested by dry filtration onto pre-weighed membranes with  $0.2\mu\text{m}$  pore-size, and rinsed of  
312 salt with a minimal amount of deionised water (adjusted to pH 7). Coccolithophore size and  
313 concentration were obtained using a Beckman Z2 Coulter Counter (see[39] for description of  
314 Coulter principle). Coccusphere and cell size were measured three times each respectively pre-

315 and post-decalcification both morning and evening on the harvest day and the preceding day.  
316 Cells were decalcified by reducing the pH of the suspension with HCl addition to 5.0 with for  
317 around 20 minutes. The Coulter counter was calibrated to use ESAW + K/2 medium as an  
318 electrolyte, and for use with the acidified electrolyte, to accommodate for the difference in ionic  
319 strength. Cell division is synchronized under the light/dark cycle and cell size was assumed  
320 to increase linearly throughout the day[40]. By measuring cell and coccosphere size morning  
321 and evening, the bias introduced due to the time of day of measurement can be removed by  
322 interpolation to the same time of day. This also removes the daily variation in the slope of  
323 Eq. 3 which is a function of cell size[40] and number of coccoliths per cell. Culture health was  
324 monitored by cell counts and microscope inspection on alternate days. Molar PIC and POC were  
325 measured with a *Rock Eval analyser*, which is preferable to making assumptions about carbon  
326 density of biogenic material. An aliquot of culture residue was bleached with dilute sodium  
327 hyperchlorite solution (4% available chlorine for 20 minutes) to remove the organic matter, and  
328 washed three times in deionised water to remove the bleach. The resultant "pseudo-sediment"  
329 was subsequently analysed using the computational software, SYRACO[41, 42, 20].

### 330 Stable isotope and Mg/Ca measurements on planktonic foraminifera

331 Paired stable isotope ( $\delta^{18}O$  and  $\delta^{13}C$ ) and Mg/Ca analyses were performed on typically 60 in-  
332 dividual shells of *Globigerina inflata* (MIS 15-13) and *Globigerina bulloides* (MIS 7-5), picked  
333 from the 300-355  $\mu\text{m}$  size fraction. In figure 2, the alternative filled points are temperatures  
334 inferred from the 250-300  $\mu$  size fraction, which captures the glacial termination more clearly  
335 than the larger fraction, but has little effect on  $[\text{CO}_2(\text{aq})]$ . Prior to isotopic analyses, samples  
336 were crushed, cleaned in 3% hydrogen peroxide solution to remove any possible organic con-  
337 taminants, rinsed with acetone and dried overnight in an oven at 60°C. Measurements of the  
338 isotopic composition of carbon dioxide, released from the foraminiferal carbonate using a MUL-  
339 TIPREP system, were performed on a VG SIRA mass spectrometer at the Univ. Cambridge.  
340 Calibration to the Vienna Peedee Belemnite standard was through the NBS19 standard[43],

341 and the analytical precision was better than 0.08 ‰ for  $\delta^{18}O$  and 0.06 ‰ for  $\delta^{13}C$ . For Mg/Ca  
342 measurements, samples were prepared following the cleaning procedure described by Barker et  
343 al. (2003)[44]. Analyses were performed on a Varian Vista Pro Inductively Coupled Plasma  
344 Optical Emission Spectrometer (ICP-OES) and a Perkin Elmer Elan DRCII quadrupole based  
345 Inductively Coupled Plasma - Mass Spectrometer (ICP-MS) at the Univ. Cambridge, following  
346 established procedures[45, 46]. Precision for measured Mg/Ca ratios determined from replicate  
347 runs of a standard solution containing Mg/Ca = 1.3 mmol/mol was 0.46%. Accuracy of Mg/Ca  
348 determinations was confirmed by interlaboratory studies of foraminifera and carbonate reference  
349 materials[47, 48].

## 350 Analytical methods

351 Proxy-derivation regressions are y-on-x ordinary least squares regressions on the mean of a fitted  
352 gaussian on log-transformed data. An ordinary least squares regression was chosen because the  
353 uncertainty in the y-axis variable far exceeds that of the x-axis variable. These histograms are  
354 given in the supplementary material. Volume-predicted PIC:POC has a highly significant rela-  
355 tionship with measured PIC:POC ( $p < 0.001$ , F-statistic = 123 on 1 and 28 DF), and an excellent  
356 linear fit ( $R_{adj}^2 = 0.81$ ). The coccolith aspect ratio (AR = coccolith thickness /  $\sqrt{\text{coccolith area}}$ )  
357 is also a highly significant predictor of PIC:POC ( $p < 0.001$ , F-statistic = 79 on 1 and 30 DF)  
358 and a good linear fit ( $R_{adj}^2 = 0.71$ ).

## 359 Down-core

360 ODP site 1123 (Expedition 181) is located on Chatham rise, east of New Zealand in the south-  
361 ernmost Pacific ( $41^{\circ}47.2'S$ ,  $171^{\circ}29.9'W$ , 3290m water depth. The sediment age model for site  
362 ODP site 1123 is based on a correlation with the orbitally tuned benthic oxygen isotope stack  
363 of LR04[49, 50]. Reconstructed sea surface temperature (SST) estimates are based on Mg/Ca  
364 ratios of planktic forams using species-specific published calibrations. Temperatures across the  
365 MIS 7-5 interval were inferred from the *G.bulloides* 250-300  $\mu\text{m}$  and 300-355  $\mu\text{m}$  size fractions



366 using the equation:  $Mg/Ca = 0.47 \exp 1.08T$ , and those across MIS 15-13 using the equation:  
367  $Mg/Ca = 0.299 \exp 0.090T$  [51].  $[CO_2(aq)]$  is estimated from global  $CO_2$  mixing ratios of an  
368 assumed well mixed atmosphere from Vostok and Dome C Antarctic ice cores (Fig. S2; com-  
369 piled by [52]), using the *seacarb* package in R [58], with dissolution assumed to be controlled only  
370 by SST at a constant salinity of 35. EDC3 gas age was converted to LR04 using a published  
371 conversion [53]. Carbon isotopic composition of planktic forams were used as a rough proxy for  
372 relative nutrient ( $\sim$  phosphate) availability corrected for the effect of temperature, using the  
373 relationships of [54]. See figure S3 for time series. Smear slides were prepared using a trial and  
374 error approach to attain the optimum coccolith density. SYRACO analysis was carried out in  
375 CEREGE.

## 376 SYRACO

377 In SYRACO analysis, all objects present in the field of view are individually segmented (coc-  
378 coliths and debris). Secondly the outline of segmented coccoliths is optimized for morphometric  
379 measurements. The threshold is computed as the average between the mode of the pixels val-  
380 ues (corresponding to the dark background) and the mean of the same segmented image. This  
381 threshold value is reproducible and works with very different taxa (F. profunda, E. huxleyi and  
382 U. sibogae to a Sphenolithus and Chiasmolithus etc.). For all images, thresholds and parameters  
383 are measured the same way. The area corresponds to the number of the isolated pixels (minus  
384 the central are if it exists) multiplied by the scale (area of 1 pixel =  $0.0036 \mu m^2$ ). The mass  
385 and the thickness is measured according to a published protocol [20].

## 386 Effect of PIC:POC on carbonate chemistry

387 Photosynthesis (net photosynthesis = photosynthesis - respiration) removes one mole of DIC from  
388 surface ocean seawater and adds 15/106 moles of alkalinity [55]) for each mole of POC produced.  
389 Calcification removes 1 mole of DIC and 2 moles of alkalinity for each mole of PIC produced.  
390  $[TA]$  relative to  $[DIC]$  dictates the buffering capacity of the surface ocean for  $CO_2$ .  $[CO_2]$  and

391 pH were calculated across DIC and TA values at salinity = 35, temperature = 25°C, assuming  
392 zero concentration of phosphate and silicate and zero hydrostatic pressure, and published values  
393 for the first and second dissociation constants of carbonic acid[56].

## 394 References

## 395 References

- 396 [1] Milliman, J. Production and accumulation of calcium carbonate in the ocean: budget of a  
397 nonsteady state. *Global Biogeochemical Cycles* **7**, 927–957 (1993).
- 398 [2] Bown, P. R. (ed.) *Calcareous nannofossil biostratigraphy*, (Chapman & Hall, Cambridge,  
399 1998).
- 400 [3] Rost, B. & Riebesell, U. Coccolithophores and the biological pump: responses to environ-  
401 mental changes. In Theirstein, H. & Young, J. (eds.) *Coccolithophores: From molecular*  
402 *processes to global impacts*, 99–124 (Springer, 2004).
- 403 [4] Boyd, P. W. & Doney, S. C. The Impact of Climate Change and Feedback Processes on the  
404 Ocean Carbon Cycle. In Fasham, M. J. (ed.) *Ocean Biogeochemistry*, chap. The Impact,  
405 157–193 (Springer, 2003).
- 406 [5] Riebesell, U. *et al.* Reduced calcification of marine plankton in response to increased atmo-  
407 spheric CO<sub>2</sub>. *Nature* **407**, 2–5 (2000).
- 408 [6] Zondervan, I., Zeebe, R., Rost, B. & Riebesell, U. Decreasing marine biogenic calcification:  
409 A negative feedback. *Global Biogeochemical Cycles* **15**, 507–516 (2001).
- 410 [7] Sett, S. *et al.* Temperature Modulates Coccolithophorid Sensitivity of Growth, Photosyn-  
411 thesis and Calcification to Increasing Seawater pCO<sub>2</sub>. *PloS one* **9**, e88308 (2014).
- 412 [8] Bach, L. T., Riebesell, U., Gutowska, M. a., Federwisch, L. & Schulz, K. G. A uni-  
413 fying concept of coccolithophore sensitivity to changing carbonate chemistry embedded  
414 in an ecological framework. *Progress in Oceanography* **135**, 125–138 (2015). URL  
415 <http://linkinghub.elsevier.com/retrieve/pii/S0079661115000725>.

- 416 [9] Bendif, E. M. *et al.* Genetic delineation between and within the widespread coccolithophore  
417 morpho-species *Emiliana huxleyi* and *Gephyrocapsa oceanica* (Haptophyta). *Journal of*  
418 *Phycology* **50**, 140–148 (2014). URL <http://doi.wiley.com/10.1111/jpy.12147>.
- 419 [10] von Dassow P, J. U. *et al.* Loss of sex in open oceans accounts for genome variability in a  
420 cosmopolitan phytoplankton. *ISME (in press)* .
- 421 [11] Irie, T., Bessho, K., Findlay, H. S. & Calosi, P. Increasing costs due to ocean acidifica-  
422 tion drives phytoplankton to be more heavily calcified: optimal growth strategy of coccol-  
423 ithophores. *PloS one* **5**, e13436 (2010).
- 424 [12] Schlüter, L. *et al.* Adaptation of a globally important coccolithophore to ocean warming  
425 and acidification. *Nature Climate Change* (2014).
- 426 [13] Lohbeck, K. T., Riebesell, U. & Reusch, T. B. H. Adaptive evolution of a key phytoplankton  
427 species to ocean acidification. *Nature Geoscience* **5**, 346–351 (2012).
- 428 [14] Beaufort, L. *et al.* Sensitivity of coccolithophores to carbonate chemistry and ocean acidi-  
429 fication. *Nature* **476**, 80–83 (2011).
- 430 [15] Horigome, M. T. *et al.* Environmental controls on the *Emiliana huxleyi* calcite mass.  
431 *Biogeosciences* **11**, 2295–2308 (2014).
- 432 [16] Iglesias-Rodriguez, M. D. *et al.* Phytoplankton calcification in a high-CO<sub>2</sub> world. *Science*  
433 (*New York, N.Y.*) **320**, 336–40 (2008).
- 434 [17] Halloran, P. R., Hall, I. R., Colmenero-Hidalgo, E. & Rickaby, R. E. M. Evidence for a multi-  
435 species coccolith volume change over the past two centuries: understanding a potential ocean  
436 acidification response. *Biogeosciences* **5**, 1651–1655 (2008).
- 437 [18] Meier, K., Berger, C. & Kinkel, H. Increasing coccolith calcification during CO<sub>2</sub> rise of the  
438 penultimate deglaciation (Termination II). *Marine Micropaleontology* **112**, 1–12 (2014).

- 439 [19] Meier, K. J. S., Beaufort, L., Heussner, S. & Ziveri, P. The role of ocean acidification in  
440 *Emiliana huxleyi* coccolith thinning in the Mediterranean Sea. *Biogeosciences* **11**, 2857–  
441 2869 (2014).
- 442 [20] Beaufort, L., Barbarin, N. & Gally, Y. Optical measurements to determine the thickness of  
443 calcite crystals and the mass of thin carbonate particles such as coccoliths. *Nature protocols*  
444 **9**, 633–42 (2014).
- 445 [21] Henderiks, J. Coccolithophore size rules ? Reconstructing ancient cell geometry and cellular  
446 calcite quota from fossil coccoliths. *Marine Micropaleontology* **67**, 143–154 (2008).
- 447 [22] Crundwell, M., Scott, G., Naish, T. & Carter, L. Glacial?interglacial ocean climate vari-  
448 ability from planktonic foraminifera during the Mid-Pleistocene transition in the temperate  
449 Southwest Pacific, ODP Site 1123. *Palaeogeography, Palaeoclimatology, Palaeoecology* **260**,  
450 202–229 (2008).
- 451 [23] Young. Neogene. In Bown, P. R. (ed.) *Calcareous Nannofossil Biostratigraphy*, 225–265  
452 (Springer, 1998).
- 453 [24] Rickaby, R. E. M., Henderiks, J. & Young, J. N. Perturbing phytoplankton: response and  
454 isotopic fractionation with changing carbonate chemistry in two coccolithophore species.  
455 *Climate of the Past* **6**, 771–785 (2010).
- 456 [25] Müller, M. N., Schulz, K. G. & Riebesell, U. Effects of long-term high CO<sub>2</sub> exposure on  
457 two species of coccolithophores. *Biogeosciences* **7**, 1109–1116 (2010).
- 458 [26] Bach, L. T., Riebesell, U. & Georg Schulz, K. Distinguishing between the effects of ocean  
459 acidification and ocean carbonation in the coccolithophore *Emiliana huxleyi*. *Limnology*  
460 *and Oceanography* **56**, 2040–2050 (2011).
- 461 [27] McClelland, H., Taylor, P., O’Dea, A. & Okamura, B. Revising and refining the bryozoan zs-  
462 MART seasonality proxy. *Palaeogeography, Palaeoclimatology, Palaeoecology* **410**, 412–420  
463 (2014).

- 464 [28] O’Dea, S. a. *et al.* Coccolithophore calcification response to past ocean acid-  
465 ification and climate change. *Nature communications* **5**, 5363 (2014). URL  
466 <http://www.ncbi.nlm.nih.gov/pubmed/25399967>.
- 467 [29] Gibbs, S. J. *et al.* Species-specific growth response of coccolithophores to Palaeocene?Eocene  
468 environmental change. *Nature Geoscience* **6**, 1–5 (2013).
- 469 [30] Raven, J. & Crawford, K. Environmental controls on coccolithophore cal-  
470 cification. *Marine Ecology Progress Series* **470**, 137–166 (2012). URL  
471 <http://www.int-res.com/abstracts/meps/v470/p137-166/>.
- 472 [31] de Vargas, C., Sáez, A., Medlin, L. & Thierstein, H. Super-species in the calcareous plankton.  
473 In *Coccolithophores: From molecular processes to global impacts* (2004).
- 474 [32] Riebesell, U., Wolf-Gladrow, D. A. & Smetacek, V. Carbon dioxide limitation of marine  
475 phytoplankton growth rates. *Nature* **361**, 249–251 (1993).
- 476 [33] Ziveri, P., de Bernardi, B., Baumann, K.-H., Stoll, H. M. & Mortyn, P. G. Sinking of  
477 coccolith carbonate and potential contribution to organic carbon ballasting in the deep  
478 ocean. *Deep Sea Research Part II: Topical Studies in Oceanography* **54**, 659–675 (2007).
- 479 [34] Sarmiento, J. & Gruber, N. *Ocean biogeochemical dynamics* (2013). URL  
480 <http://books.google.com/books?hl=en&lr=&id=QWUeAAAAQBAJ&oi=fnd&pg=PP1&dq=Ocean+biogeochemical>
- 481 [35] Moore, C. M. *et al.* Processes and patterns of oceanic nutrient limitation. *Nature Geoscience*  
482 **6**, 701–710 (2013). URL <http://www.nature.com/doi/10.1038/ngeo1765>.
- 483 [36] Loubere, P., Siedlecki, S. a. & Bradtmiller, L. I. Organic carbon and carbonate fluxes: Links  
484 to climate change. *Deep Sea Research Part II: Topical Studies in Oceanography* **54**, 437–446  
485 (2007).
- 486 [37] Berges, J. a., Franklin, D. J. & Harrison, P. J. Evolution of an Artificial Seawater Medium:  
487 Improvements in Enriched Seawater, Artificial Water Over the Last Two Decades. *Journal*  
488 *of Phycology* **37**, 1138–1145 (2001).

- 489 [38] Keller, M., Selvin, R., Claus, W. & Guillard, R. R. L. Media for the culture of oceanic  
490 ultraphytoplankton. *Journal of Phycology* **23**, 633–638 (1987).
- 491 [39] Beckmann Coulter. The Coulter Principle.
- 492 [40] Müller, M., Antia, A. & LaRoche, J. Influence of cell cycle phase on calcification in the  
493 coccolithophore *Emiliana huxleyi*. *Limnology and Oceanography* **53**, 506–512 (2008).
- 494 [41] Beaufort, L. & Dollfus, D. Automatic recognition of coccoliths by dynamical neural net-  
495 works. *Marine Micropaleontology* **51**, 57–73 (2004).
- 496 [42] Beaufort, L. Weight estimates of coccoliths using the optical properties (birefringence) of  
497 calcite. *Micropaleontology* **51**, 289–297 (2005).
- 498 [43] Coplen, T. Reporting of stable hydrogen, carbon, and oxygen isotopic abundances. *Geother-*  
499 *mics* **66**, 273–276 (1995).
- 500 [44] Barker, S., Greaves, M. & Elderfield, H. A study of cleaning procedures used for foraminiferal  
501 Mg/Ca paleothermometry. *Geochemistry, Geophysics, Geosystems* **4** (2003).
- 502 [45] de Villiers, S., Greaves, M. J. & Elderfield, H. An intensity ratio calibration method for the  
503 accurate determination of Mg/Ca and Sr/Ca of marine carbonates by ICP-AES. *Geochem-*  
504 *istry, Geophysics, Geosystems* **3** (2002).
- 505 [46] Yu, J. & Day, J. Determination of multiple element/calcium ratios in foraminiferal calcite  
506 by quadrupole ICP-MS. *Geochemistry, Geophysics ...* **6** (2005).
- 507 [47] Greaves, M. & Caillon, N. Interlaboratory comparison study of calibration standards for  
508 foraminiferal Mg/Ca thermometry. *Geochemistry, ...* **9** (2008).
- 509 [48] Rosenthal, Y. *et al.* Interlaboratory comparison study of Mg/Ca and Sr/Ca measurements in  
510 planktonic foraminifera for paleoceanographic research. *Geochemistry, Geophysics, Geosys-*  
511 *tems* **5** (2004).

- 512 [49] Elderfield, H., Ferretti, P. & Greaves, M. Evolution of Ocean Temperature and Ice Volume  
513 Through the Mid-Pleistocene Climate Transition. *Science* **704** (2012).
- 514 [50] Lisiecki, L. E. A Pliocene-Pleistocene stack of 57 globally distributed  
515 benthic  $\delta^{18}\text{O}$  records. *Paleoceanography* **20**, 1–17 (2005). URL  
516 <http://www.agu.org/pubs/crossref/2005/2004PA001071.shtml>.
- 517 [51] Anand, P., Elderfield, H. & Conte, M. Calibration of Mg/Ca thermometry in planktonic  
518 foraminifera from a sediment trap time series. *Paleoceanography* **28**, 1–15 (2003). URL  
519 <http://onlinelibrary.wiley.com/doi/10.1029/2002PA000846/pdf>.
- 520 [52] Lüthi, D. *et al.* High-resolution carbon dioxide concentration record 650,000–800,000 years  
521 before present. *Nature* **453**, 379–82 (2008).
- 522 [53] Parrenin, F. *et al.* of the Past The EDC3 chronology for the EPICA Dome C ice core  
523 485–497 (2007).
- 524 [54] Lynch-Stieglitz, J., Stocker, T. F., Fairbanks, R. G. & Broecker Wallace S. The influence of  
525 air-sea exchange on the isotopic composition of oceanic carbon: Observations and modeling.  
526 *Global Biogeochemical Cycles* **9**, 653–665 (1995).
- 527 [55] Zeebe, R. & Wolf-Gladrow, D. *CO<sub>2</sub> in Seawater: Equilibrium, Kinetics, Isotopes* (Elsevier,  
528 2001).
- 529 [56] Lueker, T. J., Dickson, A. G. & Keeling, C. D. Ocean pCO<sub>2</sub> calculated from dissolved  
530 inorganic carbon, alkalinity, and equations for K<sub>1</sub> and K<sub>2</sub>: validation based on laboratory  
531 measurements of CO<sub>2</sub> in gas and seawater at equilibrium. *Marine Chemistry* **70**, 105–119  
532 (2000).
- 533 [57] Frankignoulle, M., Canon, C. & Gattuso, J. Marine calcification as a source of carbon  
534 dioxide: Positive feedback of increasing atmospheric CO<sub>2</sub>. *Limnology and Oceanography*  
535 458–462 (1994).



536 [58] Lavigne, H., Epitalon, J.-M. & Gattuso, J. Seacarb: seawater carbonate chemistry with  
537 R. R package version 3.0.. URL <http://cran.r-project.org/package=seacarb>. (2011).

## 538 **Supplementary Information**

539 This work has online supplementary material.

## 540 **Acknowledgements**

541 This study was conducted at the University of Oxford (UK) with laboratory work at CEREGE  
542 (France). HLOM was funded by PhD studentship NE/I019522/1 in association with UKOARP.  
543 REMR acknowledges NERC grant NE/H017119/1 and ERC grant SP2-GA-2008-200915. LB  
544 is grateful for financial support from EU Seventh Framework program Past4Future and from  
545 the Agence Nationale de la Recherche under project ANR-12-B06-0007 (CALHIS). We thank  
546 Marius Müller for advice on experimental design, Yves Gally for running the SYRACO software,  
547 William Hutchison for help with GMT, Jean-Charles Mazur for laboratory assistance, and to Ian  
548 Probert for providing strains. Useful comments on the manuscript were provided by El Mahdi  
549 Bendif and Jeremy S. Hoffman.

## 550 **Author Contributions**

551 HLOM conceived and executed the study; HLOM and MH designed the culture experiments;  
552 PF and MG collected the foram trace metal and isotope data from ODP site 1123; LB and NB  
553 aided with implementation of the SYRACO software. HLOM wrote the paper in discussion with  
554 REMR; LB and MH provided interpretative input and critical review of the manuscript.

## 555 **Author Information**

556 The authors declare that they have no competing financial interests. Correspondence should be  
557 addressed to H.L.O.M (harrym@wustl.edu).

## 558 Figure captions

### 559 Figure 1

560 **Size-normalisation of coccolith mass. A**, Schematic representation of coccolithophore cell,  
561 with variables defined. **B**, Regression of molar PIC:POC ratio against volumetric ratios of calcite  
562 to organic material (Eq. 1). **C**, Regression of coccosphere radius against the square root of coccolith  
563 area. The red dashed line represents an independently derived relationship between coccolith  
564 size and coccosphere size [21]. **D**, Regression of coccosphere thickness against coccolith thick-  
565 ness. **E**, Regression of molar PIC:POC ratio against coccolith aspect ratio ( $AR_L = T_L/\sqrt{A_L}$ ;  
566 Eq. 3). The dark region around each regression line represents the  $1\sigma$  confidence interval of the  
567 regression, whilst the lighter region with the dashed border represents the  $1\sigma$  prediction interval  
568 of the regression. Error bars on individual points represent the  $1\sigma$  confidence interval of each  
569 measurement. SEM images courtesy of Jeremy Young, reproduced with permission.

### 570 Figure 2

571 **Calcification response of the Noëlaerhabdaceae to environmental changes over two**  
572 **glacial-interglacial cycles. A**, benthic  $\delta^{18}O$  [49], and 5kyr interval average sea-surface tem-  
573 perature (from Mg/Ca ratios in planktic forams; filled triangles over TII represent an alternative  
574 size fraction - see methods) and  $[CO_2(aq)]$  (calculated from  $CO_2atm$ [52] and SST) at ODP site  
575 1123 in the southern Pacific Ocean. **B-D**, Coccolith morphometrics: **(B)** Mass, **(C)** Area and  
576 **(D)** Aspect ratio. Raw data are displayed as frequency-density contour plots. Points represent  
577 5 kyr averages. *Emiliania huxley*-affiliated morphotaxa ( $MG_{Emi}$ ) are hollow squares, *Gephy-*  
578 *rocapsa* spp.-affiliated morphotaxa ( $MG_{Geo}$ ) are hollow diamonds and filled circles (population  
579 mean) represent the mean of the Noëlaerhabdaceae-affiliated morphotaxa in each sample. Arith-  
580 metic coefficient of variation (CV) is a measure of the mean-independent variance of log-normal  
581 data, and is shown in grey.

582 **Figure 3**

583 **Fractional change in PIC:POC of natural heterogeneous populations over glacial ter-**  
584 **minations (left), compared with that of monoclonal strains subject to an equivalent**  
585 **CO<sub>2</sub> change in culture (right).** **A**, The response of  $MG_{Emi}$  in nature is within the range of  
586 uncertainty and of plasticity of *E. huxleyi* observed in the laboratory. **B&C**, The PIC:POC of  
587  $MG_{Geo}$  in nature increases across both terminations, which is opposite to the equivalent plastic  
588 response as inferred from culture manipulations in *G. oceanica*. Error bars represent the  $1\sigma$   
589 uncertainty, which for the down-core response, is calculated using the prediction interval of Eq.3.  
590 For culture results, details of the regressions are given in the supplementary material.

591 **Figure 4**

592 **Instantaneous effect of biomass production on seawater carbonate chemistry as a**  
593 **function of PIC:POC ( $\rho$ ).** Solid line isocontours represent  $[CO_2]$  ( $\mu\text{mol/kg}$ ), and dashed  
594 isocontours represent pH. Solid arrows represent the instantaneous effect of biogenic matter  
595 formation, and shaded arrows  $CO_2$  exchange with an atmospheric carbon pool that is large  
596 relative to the perturbed sample of seawater. Depending on its  $\rho$ , biogenic material may form an  
597 instantaneous sink or source of  $CO_2$ . For conditions typical of the modern ocean, when  $\rho$  is  $<$  or  
598  $> 1.42$ , pH initially increases or decreases respectively, and when  $\rho$  is  $<$  or  $> 1.86$   $[CO_2]$  initially  
599 decreases or increases respectively. These critical values depend on the carbonate chemistry of  
600 the surface ocean[57].