

1 **Triple oxygen and hydrogen isotopes of gypsum hydration water for**
2 **quantitative paleo-humidity reconstruction**

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19
20 **Abstract**

21 Atmospheric relative humidity (RH) is an important parameter affecting vegetation yet palaeo-
22 proxies for RH are scarce and difficult to calibrate. Here we use triple oxygen ($\delta^{17}\text{O}$ and $\delta^{18}\text{O}$)
23 and hydrogen (δD) isotopes of structurally bounded gypsum hydration water (GHW) extracted
24 from lacustrine gypsum to quantify past changes in paleo-atmospheric RH. An evaporation

25 isotope mass balance model is used together with Monte Carlo simulations to determine the
26 range of climatological conditions that simultaneously satisfy the stable isotope results of GHW,
27 and with statistically robust estimates of uncertainty. We apply this method to reconstruct the
28 isotopic composition of paleo-waters of Lake Estanya (NE Spain) and changes in atmospheric
29 RH over the last glacial termination and Holocene (from ~15 to 0.6 cal. kyrs BP). The isotopic
30 record indicates the driest conditions occurred during the Younger Dryas (YD; ~12-13 cal. kyrs
31 BP). We estimate a RH of ~40-45% during the YD, which is ~30-35% lower than today.
32 Because of the southward displacement of the Polar Front to ~42°N, it was both windier and
33 drier during the YD than the Bølling-Allerød period and Holocene. Mean atmospheric moisture
34 gradually increased from the Preboreal to Early Holocene (~11 to 8 cal. kyrs BP, 50-60%),
35 reaching 70-75% RH from ~7.5 cal. kyrs BP until present-day. We demonstrate that combining
36 hydrogen and triple oxygen isotopes in GHW provides a powerful tool for quantitative estimates
37 of past changes in relative humidity.

38

39 **Keywords:** triple oxygen isotopes, gypsum hydration water, relative humidity, lake sediments,
40 Younger Dryas, Late Glacial-Holocene transition.

41

42 **1. Introduction**

43 The presence of gypsum ($\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$) in lacustrine sediments is commonly interpreted as
44 evidence of dry climatic conditions in the past (Hodell et al., 1995, 2005; 2012; Torfstein et al.,
45 2008; Morellón et al., 2009a; Escobar et al., 2012, amongst many others). Evaporation of Ca^{2+} -
46 SO_4^{2-} -rich lake waters can lead to gypsum supersaturation under conditions of high evaporation
47 relative to precipitation (inflow). These conditions are generally accompanied by decreased input

48 of fine-grained allochthonous sediments as a result of decreasing runoff, resulting in sediments
49 that are dominantly composed of gypsum. Interbedded layers of gypsum and other “non-
50 evaporitic” facies in lakes are often attributed to alternating wet and dry conditions (e.g. Hodell
51 et al., 1995; Ortiz et al., 2006; Morellón et al., 2009a; Escobar et al., 2012; Valero-Garcés et al.,
52 2014; Li et al., 2017).

53 The isotopic composition of lake waters is sensitive to long-term changes in the
54 Evaporation/Inflow (E/I) regime and atmospheric relative humidity (RH), among other
55 parameters (Gibson et al., 2016). In addition to E/I and RH, climatic variations recorded in
56 lacustrine carbonates (i.e. $\delta^{18}\text{O}$ of authigenic carbonates) can be masked by the effect of
57 temperature on the oxygen isotopic value during carbonate precipitation (Hodell et al., 2012). In
58 contrast, structurally bounded gypsum hydration water (GHW) can be used to reconstruct the
59 isotopic value of paleo-lake waters with little to no effect of temperature. The fractionation
60 factors for oxygen and hydrogen isotopes between the free solution and GHW are largely
61 independent of temperature in the range of most lakes (e.g. 10-35°C; Gázquez et al., 2017a).
62 Thus, the oxygen and hydrogen isotopes ($\delta^{18}\text{O}$ and δD) of GHW can be used to infer the isotopic
63 composition of paleo-lake waters at the time of gypsum precipitation (Hodell et al., 2012; Grauel
64 et al., 2016; Li et al., 2017). GHW retains the isotopic values of the parent solution provided that
65 it has not been altered by post-depositional processes (e.g. exposure to temperature >50°C after
66 deposition, solution-reprecipitation, etc.). Whether the original isotopic composition of GHW has
67 been preserved or not must be evaluated on a case-by-case basis (Hodell et al., 2012; Evans et
68 al., 2015; Gázquez et al., 2017a).

69 Recent analytical developments permit precise measurements of triple oxygen isotopes
70 ($^{17}\text{O}/^{18}\text{O}/^{16}\text{O}$), and the derived parameter ^{17}O -excess (also called $\Delta^{17}\text{O}$), in natural waters (Luz

71 and Barkan, 2010; Steig et al., 2014) and GHW (Gázquez et al., 2015) with precision better than
72 $\pm 0.01\text{‰}$ (i.e. 10 per meg; $\pm 1\sigma$). This parameter is defined as:

73
$$^{17}\text{O-excess} = \ln(\delta^{17}\text{O} + 1) - 0.528 \ln(\delta^{18}\text{O} + 1) \text{ (Eq. 1)}$$

74 where:

75 $\delta^{17}\text{O}$ and $\delta^{18}\text{O}$ denote the $^{17}\text{O}/^{16}\text{O}$ and $^{18}\text{O}/^{16}\text{O}$ in water standardized to V-SMOW (Barkan and
76 Luz, 2005; Luz and Barkan, 2010; Schoenemann et al., 2013). The value of 0.528 has been
77 proposed to describe the $\delta^{17}\text{O}$ and $\delta^{18}\text{O}$ relationship in rainwater worldwide (Luz and Barkan,
78 2010). The $^{17}\text{O-excess}$ averages ~ 37 per meg in modern meteoric waters and shows lower values
79 in evaporated water (Barkan and Luz, 2010; Steig et al., 2014; Surma et al., 2015). The trajectory
80 of $\delta^{18}\text{O}$ and $^{17}\text{O-excess}$ in evaporated water is relatively insensitive to temperature and salinities
81 below 100 g/l (Barkan and Luz, 2010; Passey et al., 2014); however, it is significantly affected
82 by other parameters such as the hydrological balance of the water body and atmospheric RH
83 (Surma et al., 2015; Gázquez et al., 2017b; Herwartz et al., 2017; see Fig. 1).

84 Despite the potential of lake sediments as palaeoclimatic archives, stable isotopes in inorganic
85 and organic proxies often allow only qualitative interpretation of past hydrological changes.
86 Quantitative reconstructions from isotope proxy data, including changes in atmospheric relative
87 humidity, have been difficult to achieve and calibrate. Here we evaluate the potential of using
88 triple oxygen and hydrogen isotopes in lacustrine GHW to quantify changes in atmospheric
89 relative humidity in the past. We use a Rayleigh evaporation isotope mass balance (IMB) to
90 estimate quantitatively climatic conditions at the time of gypsum precipitation. Monte Carlo
91 simulations are used to find the most probable solution to the model and evaluate uncertainties
92 for RH. We apply this method to isotopic data ($\delta^{17}\text{O}$, $\delta^{18}\text{O}$ and δD , and derived d-excess and
93 $^{17}\text{O-excess}$) of gypsum hydration water from Lake Estanya (Southern Pre-Pyrenees, NE Spain)

94 to infer climate during the Late Glacial and Holocene periods (ca. 15 cal. kyrs BP to 0.6 cal. kyrs
95 BP). We model the isotopic values of paleo-lake Estanya under different
96 environmental/geochemical scenarios. We compare the isotopic results and derived RH values
97 with previous sedimentological and geochemical proxies in the lake sequence (Morellón et al.,
98 2009b), as well as other biological indicators such as pollen, diatoms and chironomids (Morellón
99 et al., 2011; Vegas-Vilarrùbia et al., 2013; González-Sampéris, 2017). Lastly, we discuss more
100 generally the potential application of the method to other lakes.

101 **2. Approach and model**

102 The oxygen ($\delta^{17}\text{O}$ and $\delta^{18}\text{O}$) and hydrogen (δD) isotopic composition of lake waters increase
103 with more arid conditions and higher evaporation relative to inflow (E/I). Each isotope ratio
104 follows a slightly different fractionation leading to variability in d-excess and ^{17}O -excess
105 parameters (Surma, 2015; Gibson et al., 2016; Herwartz et al., 2017; and references therein). The
106 isotopic evolution of water during evaporation (e.g. $\delta^{18}\text{O}$ vs ^{17}O -excess and $\delta^{18}\text{O}$ vs d-excess)
107 depends on the isotopic composition of the initial water (inflow), temperature, relative humidity,
108 the isotopic composition of the water vapor in equilibrium with the liquid water and the ratio of
109 water loss by evaporation (E) with respect to the inflow (I), with the remainder lost as outflow.
110 The process is described by the expression (Criss, 1999):

111

$$112 \quad {}^*R_{WS} = \frac{{}^*\alpha_{evap}^0 \cdot (1-h) \cdot {}^*R_{WI} + {}^*\alpha_{eq}^0 \cdot h \cdot E/I \cdot R_V}{E/I + {}^*\alpha_{evap}^0 \cdot (1-h) \cdot (1-E/I)} \quad (\text{Eq. 2})$$

113

114 Where $*R_{WS}$ is the isotopic ratio of the evaporated water. $*\alpha_{\text{evap}}^0$ is the effective fractionation
115 factor, calculated as a product of the equilibrium fractionation factor ($*\alpha_{\text{eq}}^0$) and the diffusive
116 fractionation factor ($*\alpha_{\text{diff}}^0$) between the liquid water and vapor. The parameter h is the relative
117 humidity of air (0 to 1). R_{WI} is the isotopic ratio of the input prior to evaporation (i.e., the inflow
118 to the lake). R_v is the isotopic ratio of the vapor and depends on the degree to which the
119 atmospheric water vapor (v_{eq}) is in equilibrium with R_{WI} (Gibson et al., 2016), where:

$$120 \quad R_v = R_{WI} * (\alpha_{\text{eq}}^0 * v_{\text{eq}}) \text{ (Eq. 3)}$$

121 E/I represents the fraction of water loss by evaporation with respect to the inflow from the
122 system (e.g. E/I =0 means no evaporation whereas E/I =1 means all the water is lost to
123 evaporation; i.e. there is no outflow). This model assumes homogeneous isotopic composition of
124 both the liquid and vapor phases.

125 Equilibrium fractionation factors for $\delta^{18}\text{O}$ and δD are well known and calculated here as a
126 function of temperature using the equations of Horita and Wesolowski (1994). $\alpha^{17}\text{O}_{\text{eq}}^0$ is
127 calculated as $\alpha^{17}\text{O}_{\text{eq}}^0 = \alpha^{18}\text{O}_{\text{eq}}^0 \theta$, where θ is 0.529 (Barkan and Luz, 2005). Kinetic fractionation
128 during evaporation under natural conditions is not as strongly constrained as equilibrium
129 fractionation. Here we use a combination of natural and laboratory experiments to calculate α_{diff}^0
130 (Landais et al., 2006; Barkan and Luz, 2007, Luz et al., 2009). First, $\alpha^{18}\text{O}_{\text{diff}}^0$ varies as a function
131 wind driven turbulence (Dongmann et al., 1974; Uemura et al., 2010; Haese et al., 2013) (see
132 discussion section) and is calculated as:

$$133 \quad \alpha^{18}\text{O}_{\text{diff}}^0 = 1.0283^w \text{ (Eq. 4)}$$

134 where w varies between 0.5 (pure turbulent mixing; $\alpha^{18}\text{O}_{\text{diff}}^0 = 1.0141$) and 1.0 (pure diffusion;
135 $\alpha^{18}\text{O}_{\text{diff}}^0 = 1.0283$).

136 αD_{diff}^0 varies as a function of $\alpha^{18}\text{O}_{\text{diff}}^0$ and temperature (T, in °C) based on experiments by Luz
137 et al. (2009), where:

138
$$\alpha D_{\text{diff}}^0 = (1.25 - 0.02 T) (\alpha^{18}\text{O}_{\text{diff}}^0 - 1) + 1 \quad (\text{Eq. 5})$$

139

140 $\alpha^{17}\text{O}_{\text{diff}}^0$ is calculated as $\alpha^{17}\text{O}_{\text{diff}}^0 = \alpha^{18}\text{O}_{\text{diff}}^0{}^\theta$, where θ is 0.5185 (Landais et al., 2006; Barkan
141 and Luz, 2007).

142 In $\delta^{18}\text{O}$ - ^{17}O -excess and $\delta^{18}\text{O}$ -d-excess space (Fig. 1), the predicted trends of waters undergoing
143 evaporation in partial equilibrium with atmospheric vapor take the form of curves. We see that
144 both ^{17}O -excess and d-excess are largely sensitive to RH during evaporation (Fig. 1A and F),
145 whereas their sensitivities to temperature are relatively small, especially for ^{17}O -excess (Fig. 1B)
146 (Landais et al., 2006; Passey et al., 2014; Surma et al., 2015; Gázquez et al., 2017b; Herwartz et
147 al., 2017). Both cross-plots are moderately sensitive to turbulences (winds) on the water surface
148 during evaporation (Fig. 1C and H) and to the isotopic composition of the atmospheric water
149 vapor (Fig. 1D and I). Also, both isotopic systems are very sensitive to the proportion of water
150 loss by evaporation with respect to the input (E/I) (Fig. 1E and J).

151 In summary, our isotopic model is based on three equations where $\delta^{17}\text{O}$, $\delta^{18}\text{O}$ and δD are known
152 (measured), four variables that can be constrained by modern estimates ($\delta^{17}\text{O}$, $\delta^{18}\text{O}$ and δD of
153 the inflow and lake temperature), two poorly constrained but minor variables (turbulence and
154 vapor-precipitation equilibrium) and two significant unknowns (E/I and h). Estimating E/I and h
155 in the past requires some assumptions about the errors in the unknown variables and an
156 understanding of how these variables co-vary and introduce uncertainty in the model results (see
157 discussion section).

158

159 Monte Carlo simulations performed in Matlab® is used in our method to find the possible model
160 solutions that satisfy the isotopic composition of the modern and paleo-lake water given the
161 uncertainty in both the paleo-environmental parameters considered and the analytical errors. The
162 approach is represented graphically in Figure S3. Briefly, a range of model inputs is selected
163 based on conservative estimates of their distributions in the modern and potential to change in
164 the past. The distribution of model inputs is uniform and thus produces a largely uniform set of
165 model solutions. The error in each model solution is then calculated relative to the mean and 1-
166 sigma standard deviation (1SD) of each individual data point. The normalized errors for $\delta^{18}\text{O}$,
167 δD , and ^{17}O -excess are then combined to arrive at a total error (if no ^{17}O -excess data exists it is
168 excluded from the total error). Only those model solutions that fall within the 1SD are then
169 selected. In the three-dimensional space of the $\delta^{18}\text{O}$, δD , ^{17}O -excess, this can be visualised as
170 selecting all the data points that fall within an ellipsoid with axes that extend to the 1SD
171 analytical error in each parameter. From the subset of simulations, the mean and range of model
172 inputs (e.g. relative humidity) that are constituent with the lake water isotopes can derived.

173

174 **3. Materials and methods**

175 Balsas de Estanya is a karstic lake complex located at the foothills of the Southern Pyrenees
176 (42°02'N, 0°32'E) at 670 m a.s.l. It is a relatively small endorheic basin of 2.45 km² (Fig. 2) that
177 comprises multiple bodies of water. The largest and deepest lake (Estanque Grande de Abajo)
178 has been studied extensively for paleoclimate and paleolimnological reconstruction (Morellón et
179 al., 2009b; 2011; and references therein) (Fig. 2 and supplementary material). Twenty-nine
180 gypsum samples were collected from the ca. 11-m long composite sequence of Lake Estanya,
181 which is comprised of a combination of cores LEG04-1A-1K and EST06-1A-1U (Morellón et

182 al., 2009b). The cores were recovered from the deepest areas of Estanque Grande de Abajo (Fig.
183 2). The age model is based on radiocarbon, ^{137}Cs and ^{210}Pb and lithostratigraphy as previously
184 described by Morellón et al., (2009b) and Vegas-Vilarrùbia et al. (2013) (see supplementary
185 material).

186 GHW was extracted by heating the powdered gypsum *in vacuo* using a bespoke offline system
187 consisting of six vacuum lines contained within a modified gas chromatography (GC) oven in the
188 Godwin Laboratory at the University of Cambridge (UK) (Gázquez et al., 2015). Oxygen ($\delta^{17}\text{O}$
189 and $\delta^{18}\text{O}$) and hydrogen (δD) isotopes of the hydration water were measured simultaneously by
190 cavity ringdown spectroscopy (CRDS) using a L2140-i Picarro water isotope analyzer (Gázquez
191 et al., 2015 and supplementary material for details). All results are reported in parts per thousand
192 (‰) relative to V-SMOW. The uncertainty of the method was $\pm 0.05\text{‰}$ for $\delta^{17}\text{O}$, $\pm 0.1\text{‰}$ for $\delta^{18}\text{O}$
193 and $\pm 0.6\text{‰}$ for δD , $\pm 0.8\text{‰}$ for d-excess and ± 8 per meg for ^{17}O -excess (1SD).

194 Additionally, the hydration water of two samples were also analyzed for $\delta^{17}\text{O}$ and $\delta^{18}\text{O}$ using a
195 modified version of the fluorination-IRMS method of Barkan and Luz (2005) at the Institute for
196 Geology and Mineralogy at the University of Cologne, Germany (Surma et al., 2015; Gázquez et
197 al., 2015; Herwartz et al., 2017). Rain (n=59) and lake waters (n=61) collected between 2001 and
198 2012 were analyzed for oxygen and hydrogen stable isotopes (Tables S2, S3, S4 and S5 in
199 supplementary material).

200 **4. Results**

201 Twenty-nine gypsum samples, ranging in age from 14.7 to 0.6 cal. kyrs BP, were analyzed for
202 stable isotopes in GHW. The $\delta^{17}\text{O}$ varies from 3.8‰ to 7.1‰, $\delta^{18}\text{O}$ from 6.3‰ to 14.9‰ and δD
203 from -26.1‰ to 3.0‰ (Table S1 and Fig. S2 in supplementary material). The lowest values

204 correspond to gypsum samples at 136 cm below lake floor; ca. 620 cal. yr BP) and the highest
205 values to gypsum at 588 cm depth; ~12 cal. kyrs BP).

206 The oxygen and hydrogen isotope composition of the parent water from which the gypsum
207 formed is calculated from GHW using recently revised fractionation factors ($\alpha_{\text{gypsum-water}}$)
208 (Gázquez et al., 2017a) that are more precise and accurate than previous values (Gonfiantini and
209 Fontes, 1963; Fontes and Gonfiantini, 1967; Hodell et al., 2012). We use $\alpha^{18}\text{O}_{\text{gypsum-water}}$ of
210 1.00355 and $\alpha\text{D}_{\text{gypsum-water}}$ of 0.979 corresponding to a temperature of 15°C, representing roughly
211 the modern mean temperature of the lake water (~12.5°C). Note that both, $\alpha^{18}\text{O}_{\text{gypsum-water}}$ and
212 $\alpha\text{D}_{\text{gypsum-water}}$ are largely unaffected by temperature in the range from 10°C to 35°C (Gázquez et
213 al., 2017a). The use of temperatures that are 10°C higher or lower changes the $\delta^{18}\text{O}$ values by
214 only $\sim\pm 0.1\text{‰}$ and δD by $\sim\pm 2\text{‰}$, which is not very significant relative to the analytical precision
215 of our method.

216 The relation between $\alpha^{17}\text{O}_{\text{gypsum-water}}$ and $\alpha^{18}\text{O}_{\text{gypsum-water}}$ is given by the parameter θ ($\alpha^{17}\text{O}_{\text{gypsum-}}$
217 $\text{water} = \alpha^{18}\text{O}_{\text{gypsum-water}}^\theta$), which has been found to be 0.5297 ± 0.0012 and does not depend on
218 temperature between 3 and 55°C (Gázquez et al., 2017a). Therefore, we use $\alpha^{17}\text{O}_{\text{gypsum-water}}$ of
219 1.00188. Using these alpha values, we found that the paleo-lake water (i.e., GHW corrected for
220 fractionation) plot on an evaporation line with slope of 3.4 (Fig. 3). This evaporation line is
221 comprised of paleo-lake waters from different ages that evaporated under different
222 environmental conditions. Therefore, the slope of this line does not convey a unique paleo-
223 hydrological significance.

224 From 14.7 to 13.3 cal. kyrs BP during the Bølling-Allerød (B-A) period, the $\delta^{18}\text{O}$ values of the
225 lake water increased gradually from 7.8‰ at 14.7 cal. kyrs BP to 10.6‰ at 13.3 cal. kyrs BP,
226 while δD increased from 12.1‰ to 21.9‰. During the same period, d-excess varied from -

227 50.4‰ to -62.6‰. The $\delta^{18}\text{O}$ and δD of the lake water shows the highest values of the entire
228 record at ca. 12 cal. kyrs BP (11.3‰ and 23.7‰, respectively) during the Younger Dryas (YD)
229 Chronozone. This time also marked the lowest d-excess values (-69‰) (Table S1). During the
230 following Preboreal-Holocene period (from 11.7 to 7.5 cal. kyrs BP), $\delta^{18}\text{O}$ and δD values
231 decreased to \sim 5.5‰ and \sim 2.4‰, respectively. Finally, the isotopic composition of the lake water
232 reached full Holocene conditions after \sim 7.5 cal. kyrs BP ($\delta^{18}\text{O}$ of $4.3\pm 0.7\text{‰}$ and δD of -
233 $1.5\pm 2.9\text{‰}$), showing similar values to modern Lake Estanya water ($\delta^{18}\text{O}$ of $3.6\pm 0.7\text{‰}$ and δD of -
234 $2.4\pm 7.1\text{‰}$). The ^{17}O -excess of paleolake water during the Holocene ranged from -63 to -46 per
235 meg, also resembling modern values (-51 per meg). More negative ^{17}O -excess values were
236 recorded during the Preboreal-Early Holocene (-103 to -94 per meg), the YD (-82 per meg) and
237 the B-A period (-67 per meg).

238

239 **5. Discussion**

240 **5.1. Reliability of GHW results**

241 Recent stable isotope studies of GHW in lakes have produced relevant paleoclimatic records that
242 closely agree with other local and regional climatic proxies (Hodell et al., 2012; Grauel et al.,
243 2016; Li et al., 2017 and the present study). Such excellent correlations undoubtedly indicate
244 that, at least in some cases, the primary isotopic composition of GHW is preserved in time.
245 Investigations on Messinian gypsum deposits (ca. 5.9 Ma) also suggest no isotopic exchange or
246 alteration of the primary isotopic signal (Evans et al. 2015). However, it must be considered that
247 in other sedimentary sequence, GHW may have undergone isotopic modification, for example by
248 dehydration/rehydration cycles as a result exposure of gypsum to temperature $>50^{\circ}\text{C}$ (e.g. burial

249 and exhumation cycles). Therefore, the reliability of stable isotopes in GHW to reconstruct the
250 isotopic composition of paleo-waters should be evaluated on a case-by-case basis.

251 There are several lines of evidence that GHW in Lake Estanya preserves its primary isotopic
252 signature. After applying fractionation factors, the values of the Mid- Late-Holocene paleolake
253 waters roughly match the modern lake water; however, the Early Holocene and Late Glacial
254 paleolake waters (~7.5 to ~15 cal. kyrs BP) show considerably more enriched values. If the
255 GHW had exchanged with sediment pore water, we would expect a relatively homogeneous
256 isotopic profile with values similar to the current lake water. Because the burial depth is shallow
257 and sediments are porous we expect any isotopic gradients in pore water to be strongly
258 attenuated by diffusion and advection with overlying lake water.

259 The isotopic changes in GHW through the late Glacial-Holocene transition and the Holocene
260 strongly correlate with major climatic changes recorded by other regional and local paleoclimate
261 archives, including several sedimentary proxies in Lake Estanya. Accordingly, we suggest the
262 hydration water of gypsum deposits in Lake Estanya reflects the isotopic composition of the
263 paleo-lake water during the latter part of the last deglaciation and Holocene.

264

265 **5.2. Determining RH from isotopic analysis of GHW**

266 The ability of GHW to record the isotopic composition of the original fluid, with little to no
267 effect of temperature, makes it a near direct proxy for the isotopic composition of paleowater.

268 The method presented here permits $\delta^{17}\text{O}$, $\delta^{18}\text{O}$ and δD , and derived d-excess and ^{17}O -excess to
269 be determined simultaneously in the same sample. The isotopic composition of paleo-lake water
270 can then be used to model the hydrologic parameters of the basin and climatic conditions at the
271 time of gypsum precipitation. Importantly, the ability of this method to reconstruct the ^{17}O -

272 excess of paleo-waters, which is dependent of RH and practically insensitive to temperature
273 during water evaporation, constitutes a powerful tool for paleo-hydrologic reconstructions.

274 When modelling the isotopic composition of paleo-lakes to fit the GHW data, several parameters
275 must be known or assumed. The uncertainty in some variables, including temperature and the
276 isotopic composition of the freshwater member, have relatively little effect on the results of the
277 model (Fig. 4 and 5B). For example, a temperature change of 3-5°C, as expected for the last
278 Glacial-Holocene transition in some regions (see section 5.3), barely affects the model results for
279 ^{17}O -excess (up to $\sim\pm 2$ per meg in a terminal lake), whereas d-excess changes by up to $\sim 3\%$ in a
280 terminal lake, when keeping all other parameters constant. Importantly, the model solution must
281 satisfy both ^{17}O -excess and d-excess of the same paleo-water.

282 The isotopic composition of rainfall varied between glacial and interglacial periods in most
283 regions, as recorded by speleothems, paleo-groundwaters and ice cores. For example, $\delta^{18}\text{O}$ of
284 freshwater in the western-Mediterranean region increased up to $\sim 1\%$, although practically no
285 change has been observed in south Iberia during the last glacial-Holocene transition (Jasechko et
286 al., 2015) (see section 4.2. in supplementary material). As seen in Fig. 5B, when keeping other
287 parameter constant, a 1% change in $\delta^{18}\text{O}$ of the input will result in uncertainties of 3 to 4.6% in
288 the modeled RH. However, it worth noting that larger change in $\delta^{18}\text{O}$ of the rainfall may be
289 expected in other regions; for example, in areas affected by monsoonal systems and large
290 variations in the “amount effect” over glacial-interglacial cycles (e.g. southern Asia; Kathayat et
291 al., 2016). Therefore, uncertainties in the isotopic composition of the freshwater member must be
292 considered as a potential source of error for quantitative paleo-humidity estimates when using
293 this method.

294 The isotopic composition of the modern atmospheric vapor and how it varies with time is
295 unknown in most regions, nor there are estimates available for this parameter in the past.
296 However, a recent study suggests that the isotopic composition of atmospheric vapor is often in
297 partial equilibrium with that of local freshwater. A degree of equilibrium of 75% seems
298 reasonable for most tropical and intertropical regions (Gibson et al., 2016). Our IMB model does
299 not reproduce the measured $\delta^{18}\text{O}$, $\delta^{17}\text{O}$ and δD ratios for any reasonable set of input parameters
300 when full equilibrium is assumed. The model is relatively sensitive to the isotopic composition
301 of the vapor, as shown in Fig 1D. This must be considered when modeling the isotopic
302 composition of lakes in coastal areas that can be affected by advection of marine air masses,
303 whose isotopic composition is in equilibrium with seawater instead of freshwater.

304 The effect of turbulence (e.g., wind) on the isotopic equilibrium between water and vapor is
305 accounted for in our model by replacing $*\alpha_{\text{diff}}$ by $(*\alpha_{\text{diff}})^w$, where the exponent 'w' is set between
306 0.5 (pure turbulence) and 1 (no wind) (Dongmann et al., 1974; Uemura et al., 2010; Haese et al.,
307 2013). The relationship between this parameter and the wind speed is not well constrained;
308 however, it is known that the proportion of $*\alpha_{\text{diff}}$ may be suppressed by turbulent flow induced
309 by wind (e.g. Horita et al. 2008). As exemplified in Fig. 4 (see section 5.3 and supplementary
310 material for details), when turbulence is not considered, the model yields d-excess and ^{17}O -
311 excess values that are systematically too low compared to the analytical data for some periods
312 (i.e. Younger Dryas; ~12 ka). This offset can be corrected by reducing the value of the exponent
313 'w' for periods that are documented to have been windier than the average.

314 The hydrologic balance of the lake (E/I) controls the isotopic composition of the water. Lakes
315 with high E/I values (i.e. lakes of dry regions) may also show high salinities due to accumulation
316 of salts in the basin. However, the salt effect on the IMB becomes significant only at >100 g/l

317 (Sofer and Gat, 1972; Criss 1999; Herwartz et al., 2017). This salinity values may be reached in
318 some hypersaline chlorine-rich lakes, for which a salinity correction would be needed (Herwartz
319 et al., 2017). Nevertheless, gypsum precipitation does not necessary occur in high-salinity
320 environments, but often takes place in freshwater lakes saturated in calcium sulfate with
321 relatively low salinities, often ~3-4 g/l (e.g. Hodell et al., 2005; Perez-Bielsa, 2013). The E/I of
322 the lake has a large impact on the IMB. The E/I in modern lakes can be asserted by a simple
323 mass balance of conservative elements in water (e.g. sodium chloride), but this parameter in the
324 past is generally unknown. Figure 5A shows that when E/I exceeds 0.75, changes in this
325 parameter barely affect the RH values derived from the model. We estimate the errors for the RH
326 to be smaller than 5% (1SD) when the lake approached terminal conditions, as required for
327 saturation in gypsum of water. In contrast, when the model is forced to E/I <0.5 the scatter of RH
328 values increases (ca. ±15%, 1SD), suggesting that the resolution of our method for RH
329 estimation is better for lake systems close to closed conditions (all the water loss by evaporation
330 and practically not outflow) than for throughflow lakes. This is because of the $\delta^{18}\text{O}$ - ^{17}O -excess
331 and $\delta^{18}\text{O}$ -d-excess trajectories of evaporated waters under different RH diverge considerably
332 when E/I approaches 1 (Fig. 1). In contrast, the isotopic trajectory of water in a throughflow lake
333 (e.g. E/I<0.5) barely differs when evaporation occurs under different conditions of RH. This
334 indicates the method described here is especially suitable for lakes in which gypsum formed
335 under arid or semiarid conditions (Surma et al., 2015).

336 In summary, when the model is forced to match both the ^{17}O -excess and d-excess of the paleo-
337 water measured in GHW and model inputs are selected based on conservative estimates and
338 appropriate errors, the derived uncertainty in RH can be as low as ±3% (1 σ). The model is
339 insensitive to temperature changes, whereas uncertainties in the isotopic composition of the

340 rainfall can have a significant effect, especially in regions where the isotopic composition of
341 rainfall is highly variable. The accuracy of the method is best when applied to lakes under
342 arid/semiarid climate ($RH < 70\%$) and elevated E/I (hydrologically closed basins). Most of these
343 conditions are met for Lake Estanya where we have applied the method to estimate RH changes
344 during the last glacial termination and Holocene.

345

346 **5.3. Application to Lake Estanya**

347 We applied the approach described above to reconstruct paleoclimate at Lake Estanya in the
348 Southern Pyrenees during the late Glacial-Holocene transition and the Holocene. We interpret
349 past changes in the isotopic composition in GHW of Lake Estanya in terms of changing RH (see
350 supplementary material for detailed rationale about the environmental parameters selected for the
351 model).

352

353 **5.3.1. Bølling-Allerød period**

354 Between ~15 and ~13 cal. kyrs BP, coinciding with the Bølling-Allerød (B-A) period, the lake
355 showed intermediate $\delta^{18}O$ and δD values compared with the later stages. This indicates a more
356 positive water balance compared to the subsequent period (i.e. 12.8-11.6 cal. kyrs BP; Younger
357 Dryas). Our model based on ^{17}O -excess and d-excess suggests that RH during the B-A period
358 was ~55-65%. This is ~10-15% less than modern conditions in the Estanya region (~70-75%).
359 This finding is consistent with comparatively lower water salinity and greater productivity in the
360 paleo-lake than during the YD, inferred from the elemental composition (XRF) of the sediments
361 and $\delta^{13}C$ of organic matter, respectively (Morellón et al., 2009b) (Fig. 6). During the B-A period,

362 a trend towards heavier $\delta^{18}\text{O}$ and δD values was recorded, reaching a relative maximum at ca.
363 13.2 cal. kyrs BP, coinciding with a cold period.

364

365

366 **5.3.2. Younger Dryas**

367 Relatively enriched $\delta^{18}\text{O}$ and δD values (11.3‰ and 23.7‰, respectively) and lower d-excess (-
368 69‰) in the lake water recorded during the YD are also in good agreement with higher E/I
369 compared to previous and later stages. Accordingly, a maximum in water salinity during the YD
370 is also evidenced by the greatest contents of S and Ca^{2-} in this section of the core (XRF data in
371 Morellón et al., 2009b). The modelled ^{17}O -excess and d-excess of the paleolake water during the
372 YD indicate that atmospheric RH decreased to 40-45% during the aridity peak at ~12 cal. kyrs
373 BP. This is ~10-15% less than during the previous B-A period and ~30-35% less compared to
374 present. These values are also consistent with recent studies of biomarkes in Lake Meerfelder
375 Maar (Germany) that suggest RH decreased by 8-15% during YD compared with the previous
376 B/A period (Rach et al. 2017). Also, these results are in good agreement with previous studies
377 suggesting the YD in NE Spain was characterized by cold and arid conditions, with particularly
378 extreme conditions in higher altitudes of the Pyrenees (González-Sampéris et al., 2006).
379 Accordingly, generally drier conditions were recorded elsewhere on the Iberian Peninsula
380 (Moreno et al., 2012). A high-resolution speleothem record from El Seso Cave (Southern
381 Pyrenees) reveals a modest cooling of 1.3°C compared with other circum-Iberian sea surface
382 temperature reconstructions (Cacho et al., 1999; Eynaud et al., 2009), and a significant decrease
383 in rainfall during the first part of the YD (12.9-12.5 cal kryrs BP) followed by a progressive
384 increase in humidity afterwards.

385 Drier conditions throughout the YD have also been inferred from other paleoclimatic sequences
386 in Iberia (Moreno et al., 2012, Garcia-Ruiz et al., 2016; González-Sampériz et al., 2017). The
387 maximum southward migration of the polar front during the YD reached 42°N (Broecker et al.,
388 1988; Lane et al., 2013), approximately the latitude of the Southern Pyrenees and Lake Estanya.
389 Marine records from the Iberian Margin found a pronounced cooling during this period, even
390 more intense than in the LGM (Eynaud et al., 2009 and references therein). Furthermore, the
391 existence of loess deposits (13-10 cal. kyrs BP) in Central Spain (Bateman and Díez-Herrero,
392 2001) also supports an increase in aridity and perhaps wind speed in Western Europe during the
393 YD (Brauer et al., 2008), as suggested by our model results.

394

395 **5.3.3. Holocene**

396 The Early Holocene period (11.7 to 7.5 cal. kyrs BP) was characterized by a decrease in $\delta^{18}\text{O}$
397 and δD values (by $\sim 9\text{‰}$ and $\sim 20\text{‰}$, respectively). This indicates a more positive water balance
398 (lower E/I) compared to the YD. Our model indicates that RH increased to 50-60% during the
399 Early Holocene (~ 7.5 -11 cal. kyrs BP). Pollen-based vegetation reconstructions indicate
400 relatively dry conditions during the transition to the Holocene marked by increasing *Juniper sp.*
401 and decreasing mesophytes (González-Sampériz et al., 2017). The isotope values reveal a
402 comparatively large increase in humidity relative to the YD, which is not reconstructed by
403 sedimentary facies and palynology but is in agreement with other paleohydrological records of
404 NE Spain and other regions from the Iberian Peninsula (Moreno et al., 2012; Morellón et al,
405 2014; González-Sampériz et al. 2017).

406 The atmospheric reorganization following the YD led to a rapid resumption of the Atlantic
407 Meridional Overturning Circulation (AMOC) and a northwards return of the polar front to 50°-

408 60°N (Lane et al., 2013). This shifted the trajectory of the westerlies north to the Iberian
409 Peninsula, and thus weakening wind intensity and increasing humidity in Southern Europe. This
410 relative increase in moisture with respect to the previous scenario, the YD, was also reflected by
411 a decrease in the salinity of the lake water (Fig. 6) in Estanya and by an expansion of *Juniper sp.*
412 population in the watershed (Vegas-Vilarrubia et al., 2013; González-Sampériz et al., 2017).
413 During the remainder of the Holocene (7.5 to 0.6 cal. kyrs BP) the isotopic values of the lake
414 water averaged around $4.3 \pm 0.7\text{‰}$ for $\delta^{18}\text{O}$ and $-1.5 \pm 2.9\text{‰}$ for δD , and showed less variability
415 than the Late Glacial owing to higher water level. The Holocene paleo-lake water values
416 recorded by gypsum are in accordance with modern $\delta^{18}\text{O}$ and δD of the lake water, which
417 indicates that environmental conditions were similar to present. During the Mid- Late-Holocene
418 (7.5 cal. kyrs BP to 0.6 cal. kyrs BP), atmospheric RH stabilized around ~70%. This value is
419 similar to the modern RH measured in the Lake Estanya region (annual mean of ~70-75%,
420 Perez-Bielsa, 2013). These results agree with previous reconstructions based on sedimentology
421 and geochemistry, which also show rather stable conditions similar to the present with short-
422 lived abrupt hydrological fluctuations and an aridification trend after 4.5 to 4 cal. kyrs BP
423 (Morellón et al., 2009b).

424

425 **6. Conclusions**

426 We propose a new proxy for quantitative estimates of paleo-humidity. Analysis of GHW permits
427 the actual isotopic composition of paleo-waters to be determined, with little to no effect of
428 temperature. We couple triple oxygen and hydrogen isotopes in hydration water of lacustrine
429 gypsum and an Isotope Mass Balance model to quantify changes in RH in the past. Using Monte
430 Carlo simulations, the RH uncertainties derived from the input parameters to our model are

431 estimated. This can be as low as 3% (1σ) when the model is forced to match both the ^{17}O -excess
432 and d-excess of the paleo-water measured in GHW.

433 We apply this method to reconstruct the isotopic composition of paleo-waters of Lake Estanya
434 (NE Spain) and changes in atmospheric RH over the Late Glacial and Holocene periods (from
435 ~15 to 0.6 cal. kyrs BP). Our results indicate RH of 40-45% during the YD and increasing to 70-
436 75% during the Mid-Late Holocene. This suggests that the mean RH in this region during the
437 past 7.5 cal. kyr BP was similar to present (RH~75%); however, the YD was characterized by
438 much drier conditions, with atmospheric RH ~30% lower than today. The southwards shift of the
439 Polar Front to ca. 42°N during the coldest phases of the YD increased wind intensity and was
440 responsible for the minimum RH during this period.

441 The consistency of the results obtained from Lake Estanya with other proxies analyzed in this
442 lake and other regional paleoclimate records, demonstrates the reliability of isotopes in gypsum
443 hydration water as a tool for quantitative paleohydrological reconstructions in lake sediments.
444 Improving the analytical precision of triple oxygen isotope measurements in waters and better
445 understanding of the various parameters included in the model will reduce the uncertainties in
446 estimated RH.

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449

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598

599 **FIGURE CAPTIONS**

600 **Figure 1.** Sensitivity of $\delta^{18}\text{O}$ - ^{17}O -excess and $\delta^{18}\text{O}$ -d-excess to different environmental
601 parameters during evaporation of a water body ($\delta^{18}\text{O}=-8\text{‰}$, $\delta\text{D}=-54\text{‰}$, ^{17}O -excess=30 per meg
602 and d-excess=10‰) in partial equilibrium with atmospheric vapor. The isotopic composition of
603 a terminal lake has been model under different conditions of relative humidity (A and F),
604 temperature (B and G), wind (C and H) and degree of equilibrium between the atmospheric
605 vapor and the freshwater member (D and I). The isotopic compositions of water pools with
606 different ratios of Evaporation/Inflow (E/I), keeping the rest of parameter constant, are also
607 represented (E and J).

608

609 **Fig. 2.** A. Location of Lake Estanya. The red dashed line indicates the limits of the surface
610 catchment. Coring sites in Estanque Pequeño de Abajo are labelled.

611 **Fig. 3.** $\delta^{18}\text{O}$ and δD of rain and spring waters (green triangles), modern water from Lake Estanya
612 (red diamonds), and gypsum hydration waters (unfilled blue circles) ranging in age from 14.7 to
613 0.6 cal. kyrs BP. The isotopic values of paleo-lake waters (filled blue circles) were inferred using
614 isotopic fractionation factors between gypsum hydration water and the free solution (Gázquez et
615 al., 2017a).

616
617 **Fig. 4.** Results of the IMB model experiments (A. $\delta^{18}\text{O}$ vs. ^{17}O -excess; B. $\delta^{18}\text{O}$ vs. d-excess).
618 The colored diamonds represent the isotopic composition of Lake Estanya during the Holocene
619 (yellow), Preboreal (green), Younger Dryas (blue) and the Bølling-Allerød (pink) periods. The
620 model (blue lines and black dots) is tuned to fit the gypsum mother water compositions.
621 Environmental conditions for the different periods were simulated using the input parameters in
622 Table 1. The grey ellipses represent the uncertainty in the model derived from the tolerance
623 given for each input parameter.

624
625 **Fig 5.** Sensitivity experiments of the derived relative humidity to other major variables:
626 Evaporation/Input and the isotopic composition of the input. In both experiments data points
627 representative of the late-Holocene (blue markers) and the Younger Dryas (red markers) are used
628 to illustrate the potential biases in the derived relative humidity estimates: **A.** The derived
629 relative humidity shows a strong positive correlation to the assumed E/I in the range of 0.4-0.7
630 and the effect is greater at lower relative humidities (e.g. the YD). The effect is much weaker at
631 higher E/I in closed lake basins, such as Lake Estanya (as well as most other systems in which

632 evaporite mineral precipitation occurs); **B.** The derived relatively humidity with respect to the
633 assumed isotopic composition of the freshwater input. Here, relatively humidity shows a small
634 positive relationship with the isotopic composition of the freshwater input, which translates into
635 an error of between 3-4.6% in relative humidity for every 1‰ change in the freshwater water
636 input.

637

638 **Fig. 6** Isotopic composition of Lake Estanya water, reconstructed atmospheric RH during the
639 Late Glacial-Holocene transition and the Holocene and comparison with other global and
640 regional paleoclimatic archives, including, from top to bottom: 1) NGRIP $\delta^{18}\text{O}$ (Rasmussen et
641 al., 2008), 2) winter (blue line) and summer (red line) insolation at 42°N, 3) Si/Al record in
642 marine core MD99-2343, offshore Minorca (Frigola et al, 2008); 4) Alboran Sea core MD95-
643 2043 Sea Surface Temperature (SST) (Cacho et al., 1999); the isotopic composition of the paleo-
644 lake water reconstructed from gypsum hydration water (panels 5, 6 and 7); 8) atmospheric RH
645 obtained from our ^{17}O -excess/d-excess model. The mean RH obtained from the ^{17}O -excess
646 model are represented as diamonds (see Fig. 4A). The RH results of the Monte Carlo simulation
647 in two scenarios are represented by the color shading. Scenario 1 (blue banding) in which all the
648 model parameters are held constant for all time periods and scenario 2 (orange banding) in which
649 the model parameters are modified in the Younger Dryas as described in the main text and
650 shown in Table 1. Previous palaeoenvironmental reconstruction of Lake Estanya based on
651 sedimentological and geochemical proxies, including 9) $\delta^{13}\text{C}$ in organic matter, 10) a paleo-
652 salinity proxy obtained from XRF analyses of the sedimentary sequence and, 11) relative lake
653 level reconstruction (0-10 stages) based on sedimentary facies (Morellón et al., 2009b).

654

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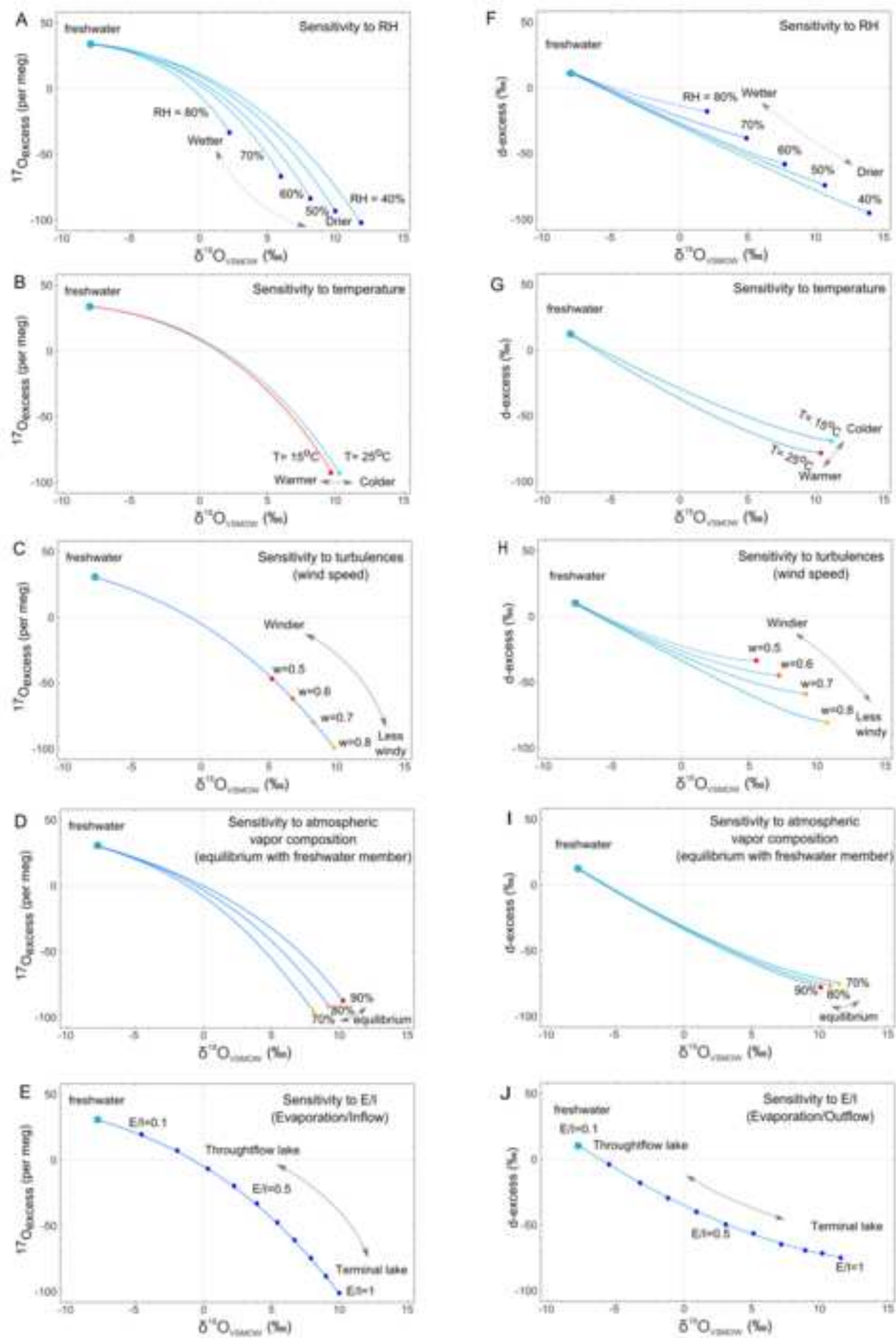


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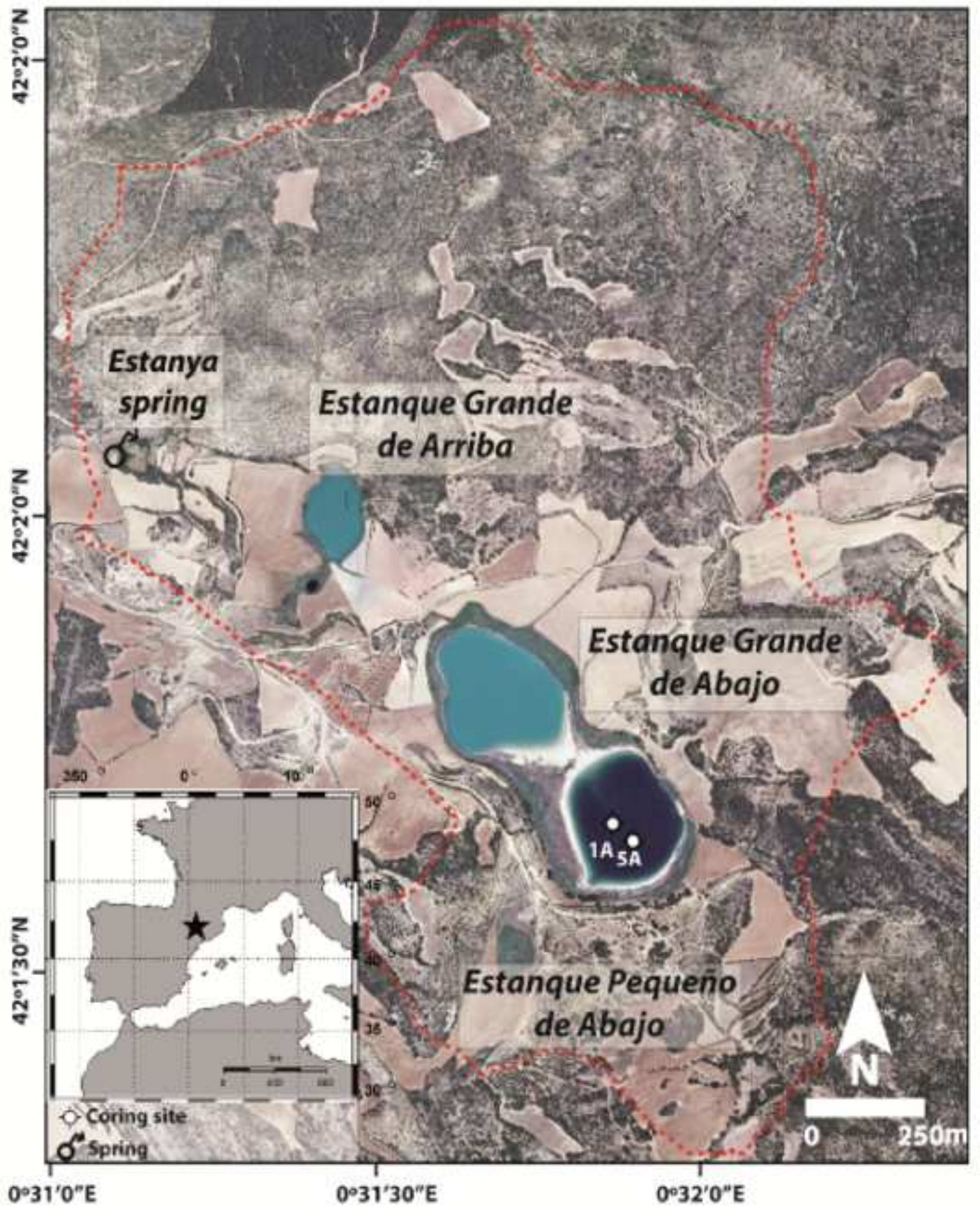


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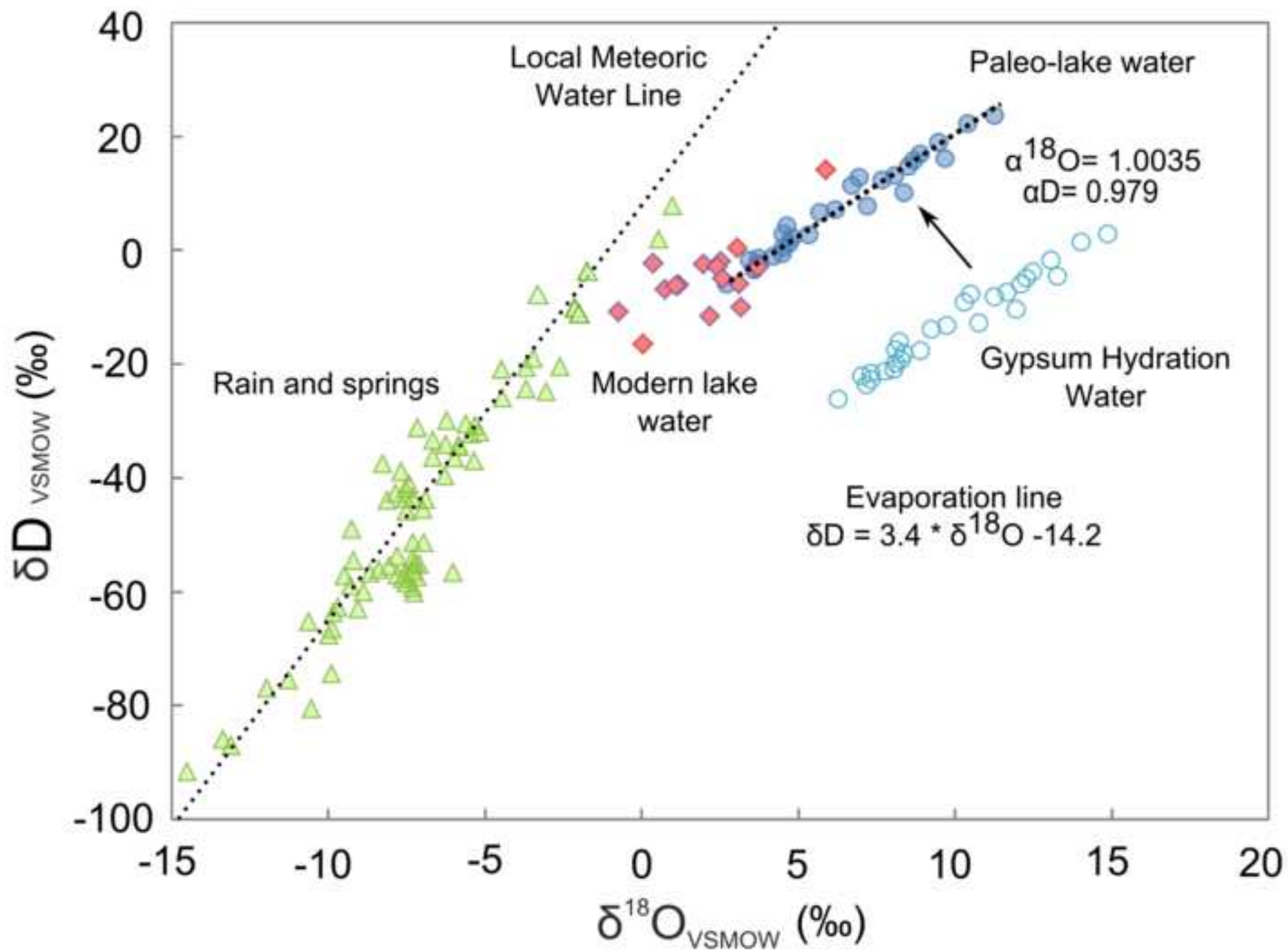


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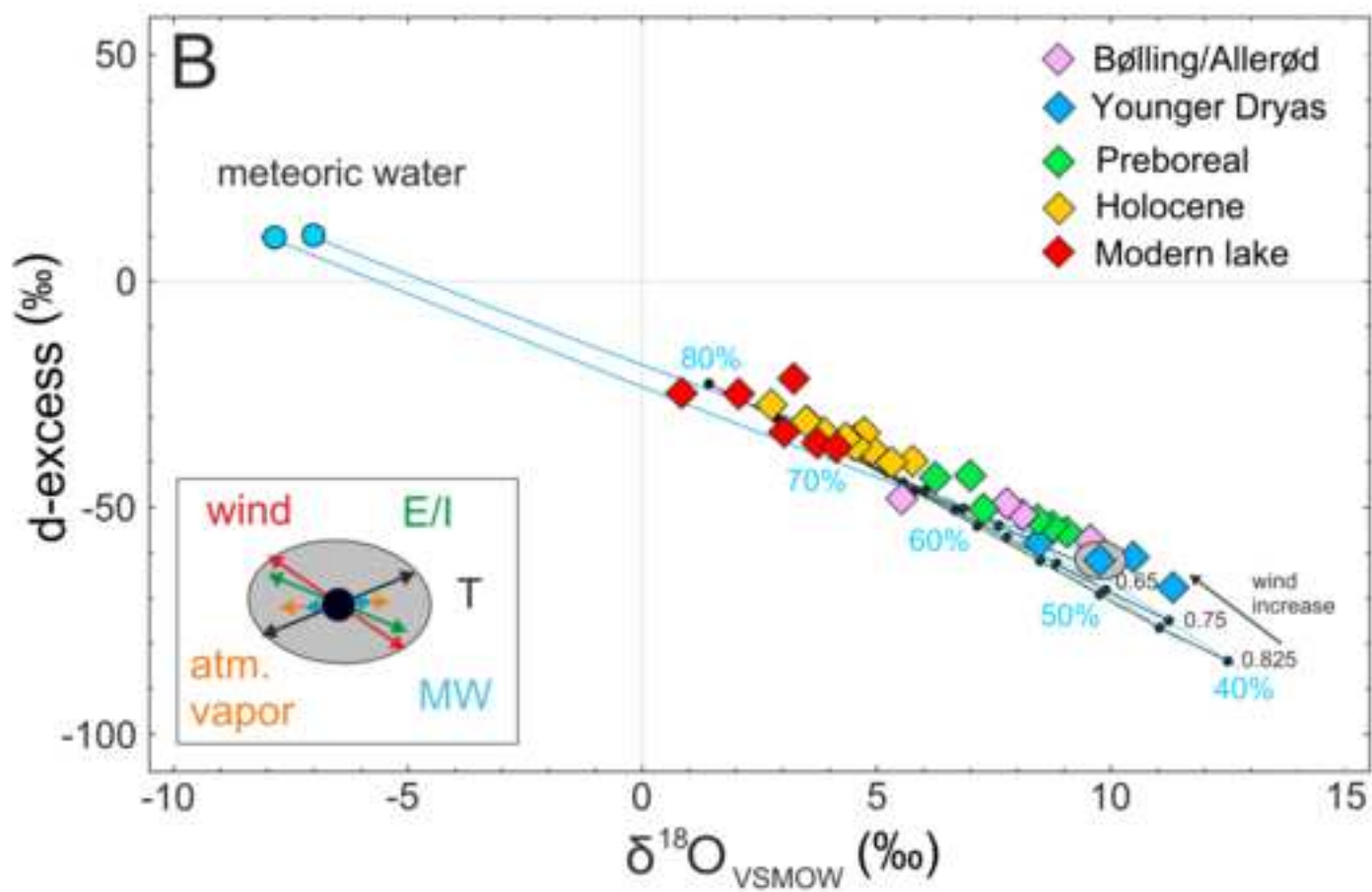
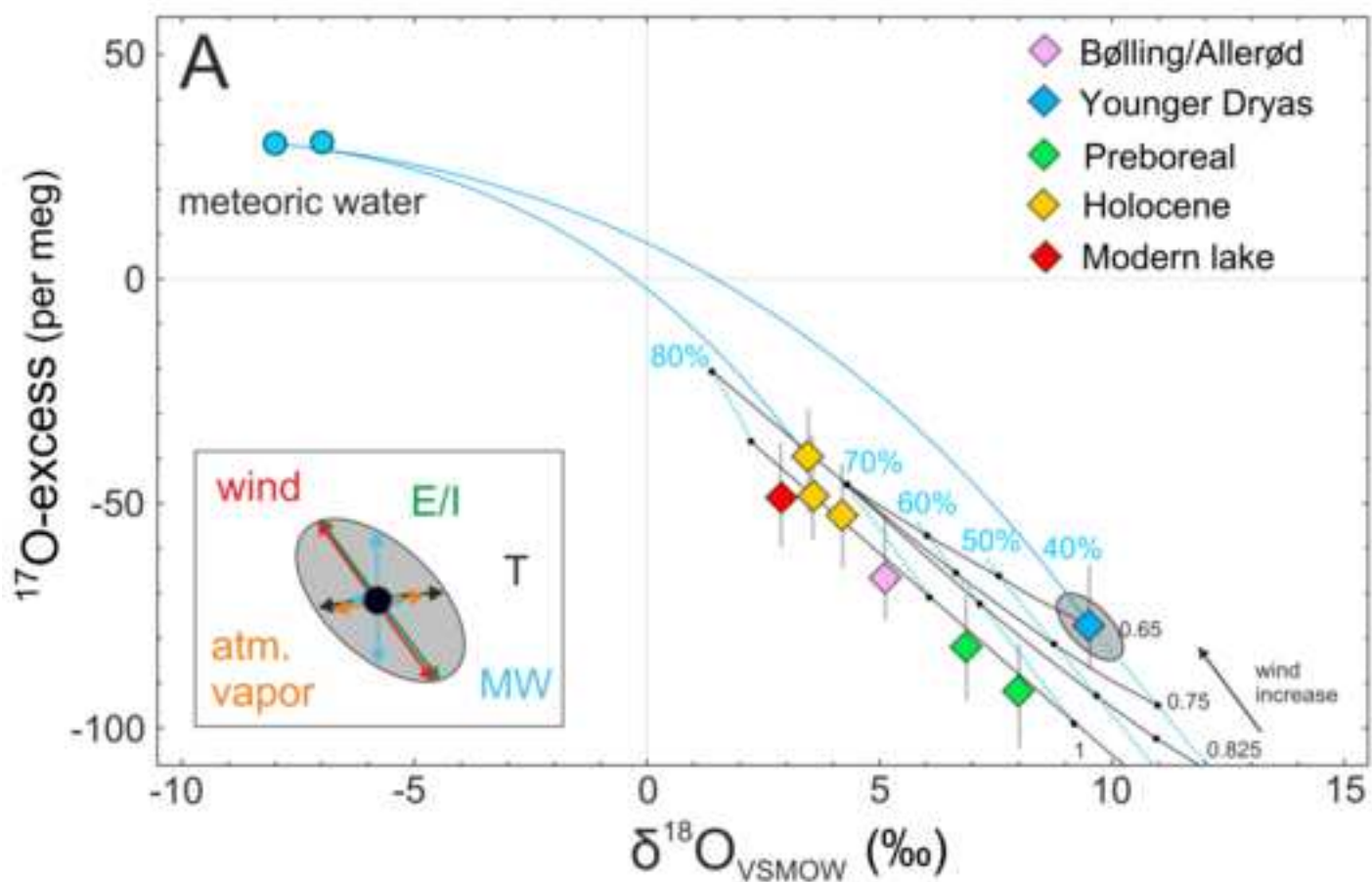


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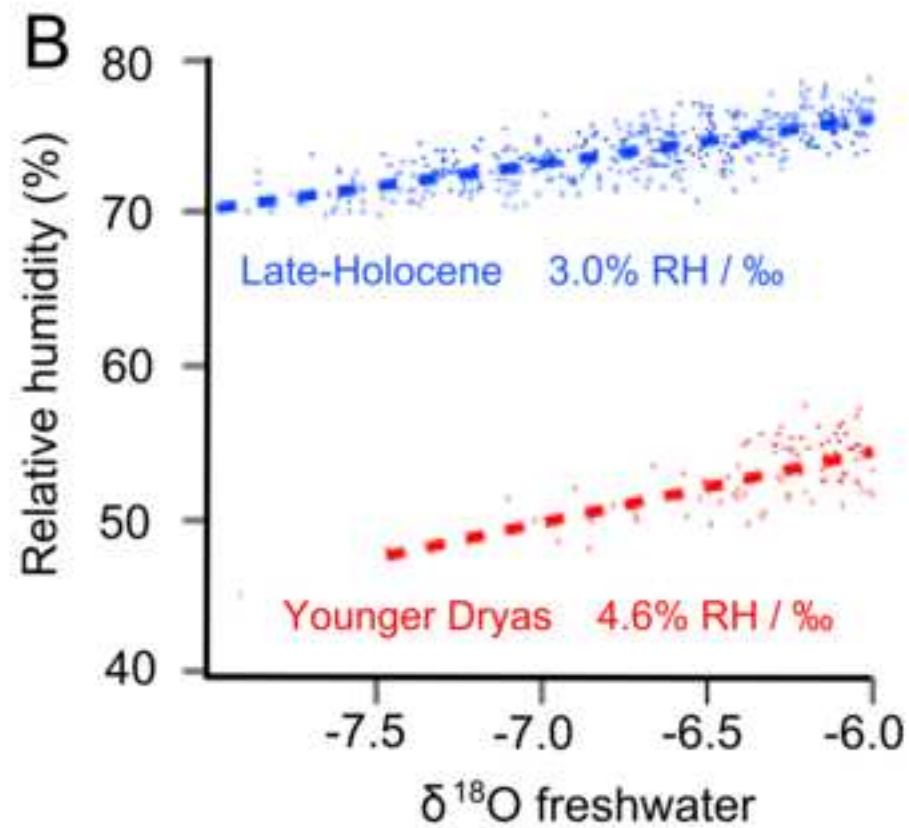
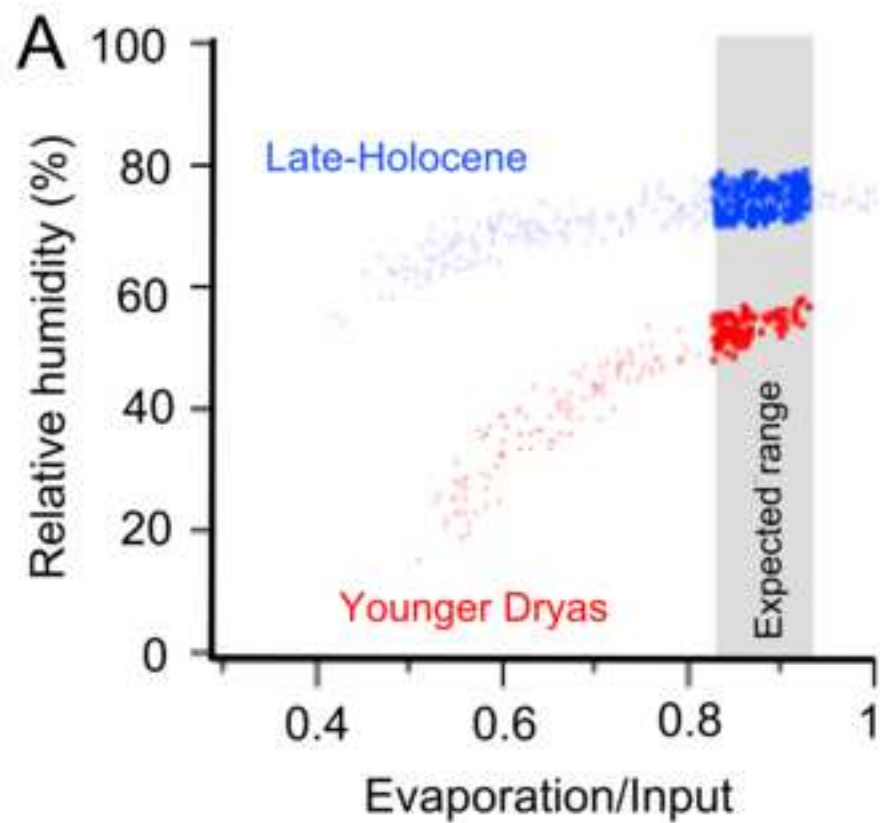
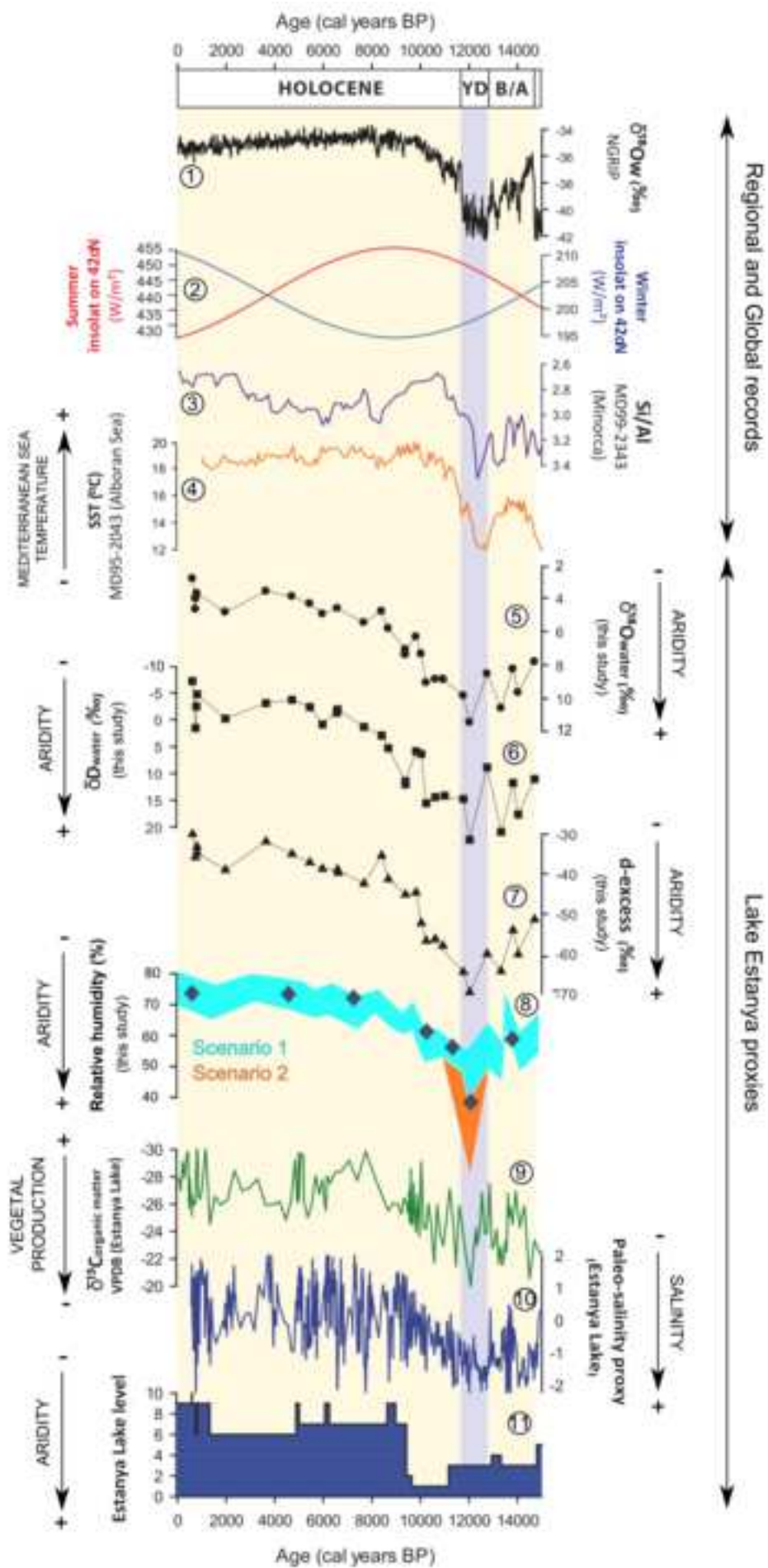


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